11819 3 Hours / 70 Marks

Seat No.

Instructions:

- (1) All Questions are *compulsory*.
- (2) Answer each next main Question on a new page.
- (3) Illustrate your answers with neat sketches wherever necessary.
- (4) Figures to the right indicate full marks.

Marks

1. Attempt any FIVE of the following:

 $5 \times 2 = 10$

- (a) Define:
 - (i) Substitution reaction
 - (ii) Elimination reaction
- (b) Name the types of organic reactions.
- (c) List the applications of methane & ethane.
- (d) Name the product in the given reaction & give its chemical formula.

$$CH_3 - CH_2 - C1 \xrightarrow{Alcoholic} KOH$$

- (e) Define power alcohol and absolute alcohol.
- (f) Distinguish between aldehyde and ketone.
- (g) Draw the structure of first four carboxylic acids in homologous series.

[1 of 4] P.T.O.

22241 [2 of 4]

2. Attempt any THREE of the following:

 $3 \times 4 = 12$

- (a) Explain Homolytic and Heterolytic fission reaction with example.
- (b) Define functional group. Write the method for determination of acid.
- (c) Observe the following reaction and identify the reactant and product.

(i)
$$CH_3 - CH_2 - I + \dots \xrightarrow{2Na, \Delta} CH_3 - CH_2 - CH_2 - CH_3$$

(ii)
$$CH_3 - CH = CH_2 + H_2 \xrightarrow{Raney \\ Nickel} \dots$$

(d) Explain the method of preparation of acetylene from calcium carbide with suitable chemical reaction.

3. Attempt any THREE of the following:

 $3 \times 4 = 12$

- (a) Give the classification of organic compound based on their structure.
- (b) Describe the method of preparing ethane by Kolbe method.
- (c) What is the reaction taking place when
 - (i) 2 Phenol is reacted with conc. HCl
 - (ii) Ethanol reacts with Acetic acid in presence of acidic medium
- (d) Explain Rosenmund reaction with suitable example.

4. Attempt any THREE of the following:

 $3 \times 4 = 12$

- (a) Given alkaline halide on heating with alcoholic solution of KOH form propylene. Predict the reactant and describe the process with chemical reaction.
- (b) Observe and complete the following:

(i)
$$CH_3 - CH - CH_3 \xrightarrow{K_2Cr_2O_7}$$
.....

(ii)
$$CH_3 - CH_2 - OH \xrightarrow{PCC} \dots$$

22241 [3 of 4]

(c) Write the IUPAC Nomenclature of the following:

(i)
$$CH_3 - CH_2 - CH - C - OH$$

 Cl

O O
$$\parallel$$
 (ii) $HO-C-CH_2-CH_2-CH_2-CH_2-CH_2-CH$

(iii)
$$CH_3 - CH - CH_2 - C - OH$$

 \mid \mid OH O

(iv)
$$CH_3 - CH - CH - CH_2 - C - OH$$

 $CH_3 - CH_3$

- (d) State the uses of acetic acid. Explain its uses in textile wet processing.
- (e) Predict the product of the following reaction. Identify name of reactants and products.

$$CH_3 - CH_2 - Br + NaCN \rightarrow \dots \qquad \xrightarrow{H_3O}^{\oplus} \dots + NH_3.$$

5. Attempt any TWO of the following:

 $2 \times 6 = 12$

- (a) Explain the mechanism of S_N^2 reaction. Draw the energy profile diagram of S_N^2 and S_N^2 reaction.
- (b) A given organic saturated compound contain four carbon atoms. One of the carbon is attached to one amino group. Predict the possible structures of compound.
- (c) Identify the compound A, B, C, D in the following series of reaction.

$$\begin{array}{c} \text{CH}_3 - \text{CH}_2 - \text{Br} \xrightarrow{\text{Alcoholic}} \dots \dots \text{(A)} \xrightarrow{\text{C}l_2/\text{CC}l_4} \dots \dots \text{(B)} \xrightarrow{\text{NaNH}_2} \dots \dots \\ \text{(C)} \xrightarrow{\text{40\% H}_2\text{SO}_4} \dots \dots \text{(D)} \end{array}$$

Write the applications of Acetylene.

22241 [4 of 4]

6. Attempt any TWO of the following:

 $2 \times 6 = 12$

- (a) Predict the reaction taking place in following cases:
 - (i) Sodium acetate is react with sodium hydroxide in presence of CaO.
 - (ii) Ethane undergo pyrolysis.
 - (iii) 1-propane react with Hydrogen Bromite in absence of $\mathrm{H_2O_2}$.
- (b) Give the preparation of ethylene glycol from ethylene and ethylene dioxide. Write the physical and chemical properties of glycol.

(c) Write any two methods of preparing ketone with suitable example.