

312336

24225

3 Hours / 70 Marks

Seat No.

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- Instructions* –
- (1) All Questions are *Compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answer with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Assume suitable data, if necessary.
 - (6) Use of Non-programmable Electronic Pocket Calculator is permissible.
 - (7) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

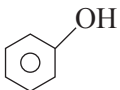
- 1. Attempt any FIVE of the following:** **10**
- a) Define acid and base by Arrhenius concept.
 - b) Define –
 - i) Osmosis
 - ii) Osmotic pressure.
 - c) State the term heat of reaction with chemical reaction.
 - d) Mention the types of organic compounds on the basis of their functional group.
 - e) State the general formula of alkene and alkyne.
 - f) Draw the structure of –
 - i) 2-methyl butane
 - ii) 2-pentene.
 - g) Define –
 - i) Melting point
 - ii) Boiling point.

P.T.O.

2. Attempt any THREE of the following: 12

- a) Explain Lewis concept of acids and bases with suitable chemical reaction.
- b) Describe the terms with one example of each –
 - i) Normality
 - ii) Gram per litre.
- c) Explain the term oxidizing and reducing agent. Give two applications of each –
 - i) KMnO_4 – Potassium permanganate
 - ii) H_2O_2 - Hydrogen peroxide.
- d) Describe the classification of organic compounds based on their functional group.

3. Attempt any THREE of the following: 12

- a) State the uses of sodium dithionate in textile wet processing.
- b) Explain four applications of organic compounds in textile industry.
- c) Write the chemical name of the following structure –
 - i) $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-COOH}$
 - ii) $\text{CH}_3\text{-CH}_2\text{-OH}$
 - iii) $\text{CH}_3\text{-CH}_2\text{-CH}=\text{CH}_2$.
 - iv) 
- d) Define crystallization. Explain with suitable example.

- 4. Attempt any THREE of the following:** **12**
- a) Elaborate the use of heat of reaction concept in textile wet processing with suitable example.
 - b) Define aliphatic hydrocarbon. Write four physical properties of aliphatic hydrocarbon.
 - c) Define distillation. Explain distillation process with neat labelled diagram.
 - d) Describe the method used to determine the boiling point of a liquid.
 - e) Explain the process of extraction theory with its principle.
- 5. Attempt any TWO of the following:** **12**
- a) Explain importance of pH in each of the following –
 - i) Scouring
 - ii) Bleaching
 - iii) Dyeing.
 - b) Describe the term “reverse osmosis.” Write two applications of reverse osmosis in textile industry.
 - c) Describe the following reactions with suitable examples –
 - i) Addition reaction
 - ii) Elimination reaction.
- 6. Attempt any TWO of the following:** **12**
- a)
 - i) Define pH and pOH. Derive relation between them.
 - ii) Explain the importance of pH in textile wet processing.
 - b) Define the term viscosity. Explain the importance of thickener viscosity on printing process.
 - c)
 - i) Differentiate aliphatic and aromatic compounds with respect to Flame test and Solubility.
 - ii) Name and write Structural formula of two aromatic and two aliphatic compounds.
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