22241

21819

3 Hours / 70 Marks

Seat No.

- Instructions (1) All Questions are Compulsory.
 - (2) Illustrate your answers with neat sketches wherever necessary.
 - (3) Figures to the right indicate full marks.
 - (4) Assume suitable data, if necessary.
 - (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. Attempt any FIVE of the following: 10

- a) Define organic compounds
- b) Write types of organic compounds
- c) State the general formula of representing alkane and alkenes.
- Identify the product and write their names.

$$2RCOO^{-}Na^{+}2H_{2}O \xrightarrow{At \text{ anode}} \frac{?}{-} + \frac{?}{At \text{ cathode}}$$

- e) Write any two industrial uses of alcohol
- Write any two points of differentiation between aldehyde and f) ketones
- Write any two physical properties of alcohol.

2. Attempt any THREE of the following:

12

- a) State four characteristic of organic compounds.
- b) Describe the method of preparation of ethane by Hurtz synthesis.
- c) What is the action of water on calcium carbide? Give balanced chemical equations
- d) Write any four industrial uses of aldehydes and ketone.

3. Attempt any THREE of the following:

12

- a) Describe the method of preparation of ethane from sodium salts of carboxylic aids.
- b) Explain homolytic bond fission with suitable examples
- c) Give any four physical properties of alkanes.
- d) How will you prepare acetic acid from Grignard reagent?

4. Attempt any THREE of the following:

12

- a) Explain elimination reaction with suitable example.
- b) Predict the products in the following reactions.

(i)
$$CH_3 - C - H + HCN \rightarrow \frac{?}{}$$

(ii)
$$CH_3 - C - CH_3 + HCN \rightarrow \frac{?}{O}$$

- c) Write any four industrial uses of aklynes.
- d) Define functional group. Write method for determines of acidic group.
- e) Write the reaction of esterification of carboxylic acid.

5. Attempt any TWO of the following:

- 12
- a) (i) Explain mechanism of SN^2 reaction draw energy for profile diagram for the same.
 - (ii) Name the following compounds.

$$\begin{array}{cccc} CH_3 & CH_3 \\ CH_3-CH_2- \begin{array}{cccc} CH & -CH_2-COOH \end{array}$$

- b) (i) Identify and name the following organic compound according to IUPAC system.
 - 1) $CH_3 CH_2 CH = CH_2$

2)
$$CH_3 - CH_3 - CH_3 - CH_3$$

3)
$$CH_3 - C - CH_2 - CH_3$$

- (ii) Classify organic compound based on their structure.
- c) (i) How will you prepare ethanol by reduction of acetaldehyde
 - (ii) Draw the structure of:
 - 1) Ethanol
 - 2) Ethylene glycol.

22241

[4]

Marks

6. Attempt any TWO of the following:

12

- a) Identify and write name of products
 - (i) $CH_2 CH_2 OH + PBr_5 \rightarrow \frac{?}{} + \frac{?}{}$
 - (ii) $3CH_3 OH + PCl_3 \rightarrow \frac{?}{} + \frac{?}{}$
 - (iii) $CH_3 CH_2 + SOCl_2 \rightarrow \frac{?}{} + \frac{?}{} + \frac{?}{}$
- b) How will you prepare acetone from isopropyl alcohol and acetylene. Explain with help of reaction.
- c) How will you prepare acetic acid from decarboxylation of malonic acid and hydrolysis of etylacetate.