

17425

11920

3 Hours / 100 Marks

Seat No.

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- Instructions* –
- (1) All Questions are *Compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Assume suitable data, if necessary.
 - (6) Use of Non-programmable Electronic Pocket Calculator is permissible.
 - (7) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.
 - (8) Use of Steam tables, logarithmic, Mollier's chart is permitted.

Marks

1. a) **Attempt any SIX of the following:** **12**
- (i) Enlist the sources of water.
 - (ii) Define hard water and soft water.
 - (iii) Define Ton of refrigeration.
 - (iv) Give the classification of boiler based on position of hot gases and water spaces with one example.
 - (v) Enlist the boiler mountings and accessories. (two each).
 - (vi) What are the application of air? (any four)
 - (vii) What are the advantages of thermic fluids over steam? (Any two).

P.T.O.

- b) **Attempt any TWO of the following:** **8**
- (i) What are the reactions that take place with hard water in lime soda process? (Any four).
 - (ii) What are the safe working properties of ideal refrigerants? (Any four).
 - (iii) What are the advantages and disadvantages of water tube boiler? (two each).
2. **Attempt any FOUR of the following:** **16**
- a) Differentiate between hot lime soda process and cold lime soda process. (Any four).
 - b) What is ecofriendly refrigerant? Give one example.
 - c) Explain with sketch fluidized bed boiler.
 - d) Draw neat diagram of sling psychrometer and explain its working.
 - e) Explain the properties of dowtherm-A.
 - f) Explain reverse osmosis process for water purification with neat sketch.
3. **Attempt any FOUR of the following:** **16**
- a) A refrigeration system operates on reverse carnot cycle. The highest temperature of the refrigerant in the system is 35°C and lowest temperature is -15°C . The capacity is to be 10 tonnes. Neglect all losses. Determine the coefficient of performance.
 - b) Explain boiler Act with respect to
 - (i) Certificate of renewal
 - (ii) Duties of inspector.
 - c) How psychrometric chart is constructed?
 - d) With neat sketch, write construction and working of forced draft cooling tower.
 - e) Give any two example of thermic fluids with their temp-range.
 - f) Draw the neat sketch of cochran boiler and label the parts.

4. Attempt any FOUR of the following:**16**

- a) Give the types of hardness of water and how the hardness of water can be measured? Define any two.
- b) What is unit of refrigeration? Give its value in SI units.
- c) What are the causes of corrosion in boiler? How corrosion due to oxygen can be prevented?
- d) Wet bulb temperature of air is 22°C and dry bulb temperature is 30°C . Find
 - (i) Dew point temperature
 - (ii) Absolute humidity by using psychrometric chart.
- e) Draw the neat sketch of thermic fluid heater and label it.
- f) Differentiate between primary and secondary refrigerants and give one example of each.

5. Attempt any FOUR of the following:**16**

- a) Describe vapour absorption refrigeration system with sketch.
- b) What are boiler mountings and explain any one with neat diagram.
- c) 50kg. of steam at 5 bar pressure has one kg of water in suspended form. What would be the condition of steam? Find the enthalpy of steam using steam table.
- d) Define humidification and relative humidity.
- e) How to prepare boiler for inspection with respect to boiler act?
- f) Give the comparison of zeolite process and Ion-exchange process of water softening.(Any four points).

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Marks

6. Attempt any TWO of the following:

16

- a) What is caustic embrittlement? What are the sources of caustic embrittlement and give two methods to prevent it.
 - b) Draw neat sketch of vapour compression refrigeration system and explain it in detail.
 - c) Find the enthalpy and entropy of 2kg of steam at a pressure of 14 bar when steam is 75% dry.
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