

22543

11920

3 Hours / 70 Marks

Seat No.

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- Instructions* – (1) All Questions are *Compulsory*.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data, if necessary.
- (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

- 1. Attempt any FIVE of the following:** **10**
- a) State the role of radiation source in analytical instruments.
- b) List any two properties of analytes used in analytical instruments.
- c) State Beer Lambert's law.
- d) Name any two applications of flame photometer.
- e) State the effect of column length in chromatography.
- f) State the principle of chromatography.
- g) State the need of pollutant monitoring.

P.T.O.

- 2. Attempt any THREE of the following:** **12**
- a) List the different filters used in analytical instruments. Explain any one filter in detail.
 - b) Draw a neat labelled diagram of glass electrode and explain its working.
 - c) List any four applications of GCMS.
 - d) Explain nitrogen oxide measurement using CO laser.
- 3. Attempt any THREE of the following:** **12**
- a) Draw a neat labelled diagram of single beam filter photometer and explain its working.
 - b) Explain the working of time of flight type mass spectrometer.
 - c) Draw a neat labelled block diagram of infrared gas analyzer. Explain its working.
 - d) Explain in brief ozone measurement using conductivity meter.
- 4. Attempt any THREE of the following:** **12**
- a) Explain the working principle of spectro photometer using prism.
 - b) In chromatography if the temperature of oven decreases, what will be its effect on retention time in chromatogram.
 - c) State the principle of thermal conductivity analyzer and explain its working.
 - d) List the types of gas pollutant and their concentration.
 - e) Explain carbon monoxide measurement using gas chromatography.

5. Attempt any TWO of the following:**12**

- a) Describe the procedure to troubleshoot NMR.
- b) Draw block diagram of LC and explain role of each block.
- c) Explain construction and working of pCO₂ electrode.

6. Attempt any TWO of the following:**12**

- a) Draw a neat labelled diagram of flame photometer. Explain the role of each element of flame photometer.
 - b) Draw labelled diagram of complete blood gas analyzer. Explain measurement of HCO₃ using same.
 - c) Explain SO₂ measurement using conductivity method.
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