

22230

11920

3 Hours / 70 Marks

Seat No.

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- Instructions* –
- (1) All Questions are *Compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Use of Non-programmable Electronic Pocket Calculator is permissible.
 - (6) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. Attempt any FIVE of the following: 10
- a) Define natural and synthetic polymer with suitable examples of each.
 - b) Define polymer and co-polymer.
 - c) Define polymerization and state its types.
 - d) Define co-ordination and chain transfer reaction.
 - e) Determine the molecular weight of $R(CH_2 - CH_2)_{200}R$
 - f) State the factors affecting the glass transition temperature.
 - g) Define photo degradation.

P.T.O.

2. Attempt any THREE of the following:**12**

- Define elastomers and fibers with suitable examples.
- Explain termination mechanism in free radical polymerization.
- Calculate the number average and weight average molecular weight of the polymer.

Molecular weight range (g/mol) x_i w_i

Mol. wt range (g/mol)	x_i	w_i
8000 - 20,000	0.05	0.02
20,000 - 32,000	0.15	0.08
32,000 - 44,000	0.21	0.17
44,000 - 56,000	0.28	0.29

- Write one reactions of each from cationic and anionic initiation reaction.

3. Attempt any THREE of the following:**12**

- Explain classification of polymers based on the structure with neat sketches.
- Compare suspension and emulsion polymerization techniques.
- Explain initiations in step polymerization.
- Explain the concept of numbers average and weight average molecular weight.

4. Attempt any THREE of the following:**12**

- Explain termination step in ring opening polymerization.
- Write any four merits and demerits of bulk polymerization.
- For number average molecular weight show that $M_n = \sum_{ni} M_i / \sum_{ni}$
- Explain the effect of plasticizers on glass transition temperature.
- Describe thermal degradation of polymer. Also state the method of prevention for the same.

5. Attempt any TWO of the following: 12

- a) Classify co-polymers with suitable examples from each category.
- b) Describe the following :
 - (i) Electrochemical polymerization.
 - (ii) Propagation mechanism in free radical polymerization.
- c) Explain the factors affecting the glass transition temperature.

6. Attempt any TWO of the following: 12

- a) Compare bulk and solution polymerization with suitable examples of both (min six points).
- b) Describe - oxidative and mechanical polymer degradation with a neat sketch.
- c) If a polymer sample has the population as follows:
 - 1 Molecule of molecular weight each = 8,000
 - 3 Molecules of molecular weight each = 7,500
 - 5 Molecules of molecular weight each = 7,000
 - 8 Molecules of molecular weight each = 6,500
 - 10 Molecules of molecular weight each = 6,000
 - 13 Molecules of molecular weight each = 5,500
 - 20 Molecules of molecular weight each = 5,000Calculate its member average molecular weight.
