

MAHARASHTRA STATE BOARD OF TECHNICAL EDUCATION (Autonomous)

(ISO/IEC - 27001 - 2005 Certified)

#### SUMMER-18 EXAMINATION Model Answer

Subject: Heat Transfer Operation

Subject code: 17560

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#### Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more Importance (Not applicable for subject English and Communication Skills.
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgement on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.



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Q No.	Answer	marks
1 A	Any three	12
1A-(i)	Thermal conductivity: It is the ability of measure of the substance to conduct	
	heat. It is the quantity of heat passing through a material of a unit thickness	2
	with a unit heat flow area in unit time when a unit temperature difference is	
	maintained across the opposite faces of the material.	
	From Fourier's law	
	Q = -kA(dT/dx)	1
	Or $k = Q.dx/(A.dT)$	
	Substituting the units	
	$k = W.m/(m^2.K)$	1
	= W/mK or J/(s.m.K)	
1A-	Film heat transfer coefficient: Film heat transfer coefficient h is defined as	2
(ii)	the quantity of heat transferred in unit time through unit area at a temperature	
	difference of $1^0$ between the surface and surrounding.	
	$1/U_o = 1/h_o + 1/h_i(D_o/D_i) + x_w/k(D_o/D_w) + R_d$	2
	$1/U_i = 1/h_i + 1/h_o(D_i/D_o) + x_w/k(D_i/D_w) + R_d$	
1A-	Stefan- Boltzman law:	
(iii)	It states that the total energy emitted (emissive power) per unit area per unit	2
	time by a black body is proportional to fourth power of its absolute	
	temperature.	
	$W_b \alpha T^4$	
	Or $W_b = \sigma T^4$	1
	Where $W_b = \text{total}$ energy emitted (emissive power) by a black body	
	$\sigma$ = Stefan Boltzman constant= 5.67*10 <sup>-8</sup> W/m <sup>2</sup> K	1



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	T = absolute temperature		
1A-	Classification of shell and tube heat exchange	jer:	1 mark
(iv)	1. Fixed tube heat exchanger		each
	2. Floating head heat exchanger		
	3. U- tube type heat exchanger		
	4. Kettle/ Reboiler type heat exchanger		
1.B	Any one		6
1B-(i)	Basis: 1 m length		
	$r_1 \!=\! 0.0525m \ r_2 \!=\! 0.0575m$		
	$r_L \!\!= (r_2 \!\!-\!\! r_1)  /  ln(r_2 \! / \! r_1) = 0.055 m$		
	$A_{L1} = 2\pi r_L L = 0.3452 \text{ m}^2$		
	$K_1 = 43.03 \text{ W/mK}$		
	$R_l = B_1 / K_l A_{Ll}$		
	= 0.005/43.03* 0.3452		1
	$= 3.37*10^{-4} \text{ K/W}$		
	$r_2 \!=\! 0.0575m \ r_3 \!=\! 0.1075m$		
	$r_L \!\!= (r_3 \!\!-\!\! r_2)  /  ln(r_3 \! / r_2) = 0.08m$		
	$A_{L2} = 2\pi r_L L = 0.5018 \text{ m}^2$		
	$K_2 = 0.07 \text{ W/mK}$		
	$R_2 = B_2 / K_2 A_{L2}$		
	= 0.05/0.07 * 0.5018		1
	= 1.423 K/W		
	$\mathbf{R}=\ \mathbf{R}_1+\mathbf{R}_2$		
	= 1.4237 K/W		1
	Temp.drop $\Delta T$ = 120 K		
	Heat loss $Q = \Delta T / R$		
	= 120 / 1.4237		1



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	$= 84.29 \text{ W}$ $Q = (T_1-T_2) / R_1 \text{ where } T_2 \text{ is the temperature at interface}$	1
	$84.29 = (423-T_2) / 3.37*10^{-4}$ $T_2 = 422.97 \text{ K}$	1
1B-	Methods of increasing the economy of an evaporator:	
(ii)	1. Using multiple effect evaporator	2
	2. Vapour recompression	
	A. Multiple effect evaporation: In this system, evaporators are arranged	
	in series so that the vapour produced in first effect is fed to the steam	
	chest of second effect as heating medium in which boiling takes place at	
	low pressure and temperature and so on.	
	B. Methods of increasing economy by vapour recompression methods	
	are:	
	1. Mechanical recompression	
	2. Thermal recompression	
	Thermal recompression: To increase the economy of single effect evaporator,	
	the principle of thermal recompression is used. Here the vapour from the	
	evaporator is compressed to increase its temperature so that it will condense at a	2
	temperature higher enough to permit its use as heating media in the same	
	evaporator. In this method, vapour is compressed by means of jet ejectors. Here	
	the high pressure steam is used to draw and compress the major part of vapours	
	from the evaporator, while the remaining part of vapours is separately	
	condensed for compensating motive steam added.	



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	Thickness of insulation Optimum Thickness Of Insulation	1
2-b	Fourier's law of conduction:	
20	It states that the rate of heat flow across an isothermal surface is proportional to	2
	the temperature gradient at the surface.	
	$\frac{dQ}{dA} = -k\frac{\delta T}{\delta n}$	1
	Q- rate of heat transfer	1
	A- Area perpendicular to heat flow	
	k- Thermal conductivity	
	T- Temperature	
2-c	Kirchhoff's Law :	
	Consider that the two bodies are kept into a furnace held at constant	2
	temperature of T K. Assume that, of the two bodies one is a black body& the	
	other is a non-black body i.e. the body having 'a' value less than one. Both the	
	bodies will eventually attain the temperature of T K & the bodies neither	
	become hotter nor cooler than the furnace. At this condition of thermal	
	equilibrium, each body absorbs and emits thermal radiation at the same rate.	
	The rate of absorption & emission for the black body will be different from that	
	of he non-black body.	
	Let the area of non-black body be $A_1$ and $A_2$ respectively. Let 'I' be the rate at	



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	which radiation falling on bodies per unit area and $E_1$ and $E_2$ be the emissive	
	powers ( emissive power is the total quantity of radiant energy emitted by a	
	body per unit area per unit time)of non-black & black body respectively.	
	At thermal equilibrium, absorption and emission rates are equal, thus,	2
	$Ia_1 A_1 = A_1 E_1$ (1.1)	
	$\therefore$ Ia <sub>1</sub> = E <sub>1</sub> (1.2)	
	And $Ia_b A_2 = A_2 E_b$ (1.3)	
	$Ia_b = E_b \qquad \dots $	
	From equation (1.1) and (1.4).we get	
	$\frac{E1}{a1} = \frac{Eb}{ab} \qquad \dots $	
	Where $a_{1,}a_{b}$ = absorptivity of non-black & black bodies respectively.	
	If we introduce a second body (non-black) then for the second non-black	
	body,we have :	
	$I A_3 a_2 = E_2 A_3$ (1.6)	
	$\therefore Ia_2 = E_2 \qquad \dots $	
	Where $a_1 = E_2$ are the absorptivity and emissive power of the second non-black	
	body.	
	Combining equations $(1.2)$ , $(1.4)$ and $(1.7)$ we get,	
	$\frac{E_1}{a_1} = \frac{E_2}{a_2} = \frac{E_3}{a_3} = E_b \qquad \dots $	
2-d	Application of finned tube heat exchanger: When the heat transfer	2
	coefficient of one of the process fluids is very low as compared to the other, the	
	overall heat transfer coefficient becomes approximately equal to the lower	
	coefficient. This reduces the capacity per unit area of the heat transfer surface	







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$A = 19.95 m^2$			
Dropwise and f	ïlmwise condensation:		2 mark
Points	Dropwise condensation	Filmwise	each fo
		condensation	any
mechanism	In case of drop-wise condensation	In case of film-wise	
	the condensate (condensed liquid)	condensation the	
	does not wet the surface and	condensed liquid wets	
	collects to grow for a while and	the surface and forms a	
	then fall from the surface, leaving	continuous film of	
	bare metal surface for further	condensate through	
	condensation.	which heat transfer	
		takes place. This	
		condensate flows down	
		due to action of gravity	
Heat transfer	Heat transfer coefficient are very	Heat transfer	
coefficient	high in case of drop-wise	coefficients are	
	condensation since the heat does	relatively very low in	
	not have to flow through film by	case of film-wise	
	conduction	condensation since the	
		heat does have to flow	
		through film by	
		conduction	
Surface type	Oily or greasy surfaces seem to	Smooth, clean surfaces	
	tend towards drop-wise	seem to tend towards	
	condensation	film-wise condensation	
Stability	Drop-wise condensation is very	Film-wise	



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		difficult to achieve an	nd unstable	condens	ation is easily	y
				obtainab	ole and stable	
	equations	If the students write e	equations for	If the stu	udents write	
		film coefficients on v	vertical and	equation	ns for film	
		horizontal surfaces m	arks should	coefficie	ents on vertica	al
		be given		and hori	zontal surface	es
				marks sł	hould be give	n
3-с	Comparison of s	quare pitch and trian	gular pitch(a	ny 4)		1.5 mark
	Squ	are pitch	Tr	iangular	pitch	each
	Permits external	cleaning of the tubes	Difficult to c	lean		
	Causes low pres	sure drop on the shell	op on the shell Causes more pressure drop			
	side fluid					
	Less no. of	tubes can be Larger no. of tubes can be		be		
	accommodated	than with triangular	accommodat	ed in a	a given she	ell
	pitch		diameter			
	Creates con	mparatively less Creates large turbulence in the shell		ell		
	turbulence		side fluid			
	Can be used for	dirty fluids also	Used for clea	ın fluid		
	Use of baffle:		I			
	1. To increa	se the rate of heat the	ransfer by inc	creasing t	he velocity a	and 1 mark
	turbulence of the shell side fluid.			each		
	2. Structural	support for the tubes a	and dampers ag	gainst vib	ration.	
4 A	Any three					12
4A-(i)	Heat transfer through single flat furnace wall :			2		
	Consider that a w	all is made of material	of thermal co	nductivity	K & is of	
	uniform thickness	s (x) & constant cross s	sectional area	(A) .Assu	me K is	



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	independent of temperature & heat losse	es to atmosphere is negligible. Hot fa	ce
	is at a temperature $T_1$ & cold face is at a	temperature $T_2$ . The direction of heat	ıt
	flow is perpendicular to the wall & T va	ries in direction of X-axis.	
	T face to the cold for a cold for	ace	
	At Steady State, there can be neither acc	cumulation nor depletion of heat with	in
	a plane wall &Q is constant along heat f	low. The ordinary use of Fourier's L	aw
	requires that the differential eqn is integ	rated over entire path from $x = 0, x =$	х.
	$\therefore Q = -K$	X AdT/dx	2
	Q dx = -1	K A dT	
	OR		
	$Q_0 J^A dx = -K$	$(A_{T1})^{T2}.dt$	
		$(A_{T1})^{T2}.dt$ $(A_{T2} - T_{1})$	
	Q.x = -1		
4A-	Q.x = - I	$X A (T_2 - T_1)$	1 mark
4A- (ii)	Q.x = - I OR $Q = K A (T_2 - T_1) / x$	$X A (T_2 - T_1)$	1 mark each
	$Q.x = -I$ OR $Q = K A (T_2 - T_1) / x$ Forward feed and backward feed arra	X A $(T_2 - T_1)$ angements: (any 4)	each







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	conductivity. Graphite being soft, these exchangers are made in cubic or	
	cylindrical blocks. In cubic exchangers, parallel holes are drilled in a solid cube	
	such that parallel holes of a particular row are at right angles to the holes of the	
	row above & below. Headers bolted to the opposite sides of the vertical faces of	
	the cube provide the flow of process fluid through the block. The headers	
	located on the remaining vertical faces direct the service fluid through the	
	exchanger in a cross flow.	
		2
4 B	Any one	6
4B-(i)	let area = $1 \text{ m}^2$	
	Thermal resistance o fire brick = $x_1/k_1 A$	1
	Thermal resistance o fire brick = $x_1/k_1 A$ R <sub>1</sub> = 0.23/1.21x1 = 0.190 k/w	1
		1
	$R_1 = 0.23/1.21 x 1 = 0.190 k/w$	1
	$R_1 = 0.23/1.21x1 = 0.190 \text{ k/w}$ Similarly $R_2 = x_2/k_2 \text{ A} = 0.075/0.121x1 = 0.62 \text{ k/w}$	1
	$R_1 = 0.23/1.21x1 = 0.190 \text{ k/w}$ Similarly $R_2 = x_2/k_2 \text{ A} = 0.075/0.121x1 = 0.62 \text{ k/w}$ $R_3 = x_3/k_3 \text{ A} = 0.089/0.865x1 = 0.103 \text{ k/w}$	
	$R_1 = 0.23/1.21x1 = 0.190 \text{ k/w}$ Similarly $R_2 = x_2/k_2 \text{ A} = 0.075/0.121x1 = 0.62 \text{ k/w}$ $R_3 = x_3/k_3 \text{ A} = 0.089/0.865x1 = 0.103 \text{ k/w}$ $R = R_1 + R_2 + R_3$	
	$R_{1} = 0.23/1.21x1 = 0.190 \text{ k/w}$ Similarly $R_{2} = x_{2}/k_{2} \text{ A} = 0.075/0.121x1 = 0.62 \text{ k/w}$ $R_{3} = x_{3}/k_{3} \text{ A} = 0.089/0.865x1 = 0.103 \text{ k/w}$ $R = R_{1} + R_{2} + R_{3}$ R = 0.913  k/w	
	$R_{1} = 0.23/1.21x1 = 0.190 \text{ k/w}$ Similarly $R_{2} = x_{2}/k_{2} \text{ A} = 0.075/0.121x1 = 0.62 \text{ k/w}$ $R_{3} = x_{3}/k_{3} \text{ A} = 0.089/0.865x1 = 0.103 \text{ k/w}$ $R = R_{1} + R_{2} + R_{3}$ R = 0.913  k/w The heat loss per unit area is $Q = \Delta T/R$	







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	pipe connecting the vapour space to the bottom of the exchanger is	2
	provided for natural circulation of a unvapourised liquid. It is provided with	
	inlet connection for feed, steam and outlet connections for vapour, thick	
	liquor, condensate etc.	
	Working:	
	In this evaporator feed enters the bottom of the tubes, gates heated by the	
	condensing steam, starts to boil part way up the tubes and the mixture of	
	vap. and liquid comes out from the top of the tubes and finally impinges at	
	high velocity on a deflector. The deflector acts both as a primary separator	
	and foam breaker. The separated liquid enters the bottom of the exchanger	
	and parts of this liquid is taken out as a product.	2
	This type of evaporator is widely used for handling of foamy, frothy	
	liquids.	
	It is typically used for the production of condensed milk and concentrating	
	black liquor in the pulp and paper industry.	
5	Any two	16
5-a	To derive $Q=UA \Delta T_{lm}$	
	Assumptions:	
	1. Overall coefficient U is constant throughout the exchanger	2
	2. Specific heats of hot and cold fluids are constant	
	3. Heat flow to and from the ambient is negligible	
	4. Flow is steady and may be parallel or counter current type	
	5. Temperatures of both the fluids are uniform over a given cross section	
	and may be represented by their bulk temperature.	
L		<u> </u>





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	$\int_{\Delta Ti} d(\Delta T) / \Delta T = - (1/(mh Cph) + 1/(mc Cpc)) U B \int_0^L dt$	x	
	$\ln (\Delta Te/\Delta Ti) = - (1/(mh Cph) + 1/(mc Cpc)) UA$		-(6)
	where $\Delta Te = T_{he} - T_{ce}$		1
	$\Delta Ti = T_{hi} - T_{ci}$		
	Now if q is the total rate of heat transfer in the heat exchan	nger, then	
	$q = m_h C p_h (T_{hi} - T_{he})$ (7)		
	= mc Cpc (T <sub>ce</sub> - T <sub>ci</sub> )(8)		
	Substituting equations (7) and (8) into equation (6),		1
	$\ln (\Delta Te/\Delta Ti) = -1/q[ (T_{hi}-T_{he}) + (T_{ce}-T_{ci})]U A$		
	$q = U A (\Delta Ti - \Delta Te) / \ln (\Delta Ti / \Delta Te)$	(9)	
	Equation (9) is the performance equation for a parallel-flo	w heat exchanger	
	$Q = U A \Delta T lm$		
	Where $\Delta Tlm = (\Delta Ti - \Delta Te) / \ln (\Delta Ti / \Delta Te)$		1
5-b	Material balance equation for single effect evaporator:	:	
	Consider that the evaporator is fed with $m_f kg/h$ of weak s	olution containin	g w <sub>1</sub>
	% solute & thick liquor is withdrawn at m' kg/h containin	g $w_2$ % solids by	1
	weight. Let $m_v$ be the kg/h of water evaporated. Then :		



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	Heat transfer to solution in evaporator by condensing steam (in absece of heat	
	losses) is utilised to heat the feed solution from Tf to T and for vaporisation of	1
	water from solution.	
	Qs = Q	
	$= m_f \operatorname{Cpf} (T - T_f) + (m_f - m') \lambda_v \dots \dots (vii)$	
	$m_s \ . \ \lambda_s = mf \ Cpf \ (T-T_f) + (m_f - m') \ \lambda_v . \ldots (viii)$	
	where $Cp_f =$ specific heat of feed solution	
	$\lambda_v =$ latent heat of evaporation from thick liquor	
	For negligible boiling point rise $\lambda v = \lambda$	
	Where $\lambda$ =latent heat of vaporisation of water at pressure in the	
	Vapour space & can be read from steam tables.	1
	Above equation (viii) becomes :	
	$m_s \lambda_s = m_f C p_f (T - T_f) + (m_f - m') \lambda \dots (ix)$	
	$m_s  \lambda_s = m_f  C p_f  (T - T_f) + m_v  \lambda(x)$	
5-c	The Sider – Tate equation is	2
	hi Di/k = 0.023 (NRe <sup>) 0.8</sup> (Npr) <sup>1/3</sup> ( $\mu/\mu w$ ) <sup>0.14</sup>	
	Substituting all the values in the equation we get	
	hi $(0.02)/0.25 = 0.023 \text{ x} (15745)^{0.8} (36)^{1/3} \text{ x} ((550 \text{ x} 10^{-6})/(900 \text{ x} 10^{-6}))^{0.14}$	2
	hi (0.02)/0.25 = 0.023 x 2278.84 x 3.3 x 0.933	
	hi (0.02)/0.25 = 161.37	2
	hi= 2017	
	Inside heat transfer coefficient = $2017 \text{ W/m}^2$ .K	2
6	Any two	16
6-а	Dimensional Analysis :	2
	It is a method of correlating a number of variables into a single equation	
	expressing an effect.	



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Dimensional analysis is a method o	f reducing the number of variables required	L L
to describe a given physical situatio	n by making use of the information implied	b
by the units of the physical quantiti	es involved. It is also known as the "theory	y
of similarity".		
Dittus – Bolter equation:		
$hD/k = 0.023[(Du\rho/\mu)^{0.8}(Cp \mu/k)^{a}$		1
where $a = 0.4$ for heating		
a=0.3 for cooling.		
where h= film heat transfer coeffici	ent	
D= diameter of pipe line		
$\mu$ = viscosity of the liquid		2
$\mu$ w= viscosity of the liquid at the wa	all surface temp	
Cp= specific heat of the liquid		
L= length of pipe.		
k= thermal conductivity		
u= velocity of flow		
The Sider – Tate equation is		
hi Di/k = 0.023 (NRe) <sup>0.8</sup> (Npr) <sup>1/3</sup> ( $\mu$	$(\mu w)^{0.14}$	1
where $h = film$ heat transfer coeffici	ent	
D= diameter of pipe line		
$\mu$ = viscosity of the liquid		
$\mu$ w= viscosity of the liquid at the wa	all surface temp	
Cp= specific heat of the liquid		2
L= length of pipe.		
k= thermal conductivity		
u= velocity of flow		
		1



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6-b	$328 \text{ K} \xrightarrow{\text{Cold fluid}} 358 \text{ K} \qquad (t_1) 328 \xrightarrow{\text{Cold fluid}}$	$\rightarrow$ 358 K (t <sub>2</sub> )	
	578 K Thermic fluid 433 K 433 K (T <sub>1</sub> )	<u>Thermic fluid</u> 5	278 K 2
	Co-current flow Counter cu	rrent flow	
	co current flow		
	$\Delta T_1 = 578 - 328 = 250 \text{ K}$		
	$\Delta T_2 = 433 - 358 = 75$ K		1
	$LMTD = \frac{\Delta T1 - \Delta T2}{\ln(\frac{\Delta T1}{\Delta T2})} = \frac{250 - 75}{\ln(\frac{250}{75})} =$	145.35K	1
	Total heat transferred $Q = U A LMTD$		1
	= 700 * 500 * 145.35		
	= 50873242.14 W or 50873	.242 kW	
	counter current flow		1
	$\Delta T_1 = 433 - 328 = 105 \text{ K}$		
	$\Delta T_2 = 578 - 358 = 220 \text{ K}$		1
	$LMTD = \frac{\Delta T1 - \Delta T2}{\ln(\frac{\Delta T1}{\Delta T2})} = \frac{105 - 220}{\ln(\frac{105}{220})} =$	= 155.48K	
	Total heat transferred $Q = U A LMTD$		1
	= 700 * 500 * 155.48		
	= 54416364.83 W or 54416	.364 kW	1
6-с	Basis: 5000 kg/hr feed is fed to the evaporator.		
	Material balance of solids:		



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Solids in feed= solids in the thick liquor		1
0.01x5000=0.4 x m'		
m'=1250kg/h.		
overall Material balance:		
kg/h feed= $kg/h$ water evaporated + $kg/h$ this	ck liquor	1
water evaporated( m <sub>v</sub> )=5000-1750=3750kg/l	h	
Energy balance is		
$m_s \lambda_s = m^* c_{pf}^* (T-T_f) + m_v \lambda_v$		1
$m_s 2162 = 5000*4.187*(373-313) + 3750$ (2)	2676-419)	
steam fed(m <sub>s</sub> )= 4495.77 kg/h		1
steam economy= kg/h water evaporated/kg/h	n steam consumed	
= 3750/4495.77= <b>0.834</b>		1
$Q = U^*A^*\Delta T$		1
4495.77*2162*1000/ 3600 = 1750 * A*(37	/3-313)	
$A = 45.38 m^2$		2