# 17312

## 16117 3 Hours / 100 Marks

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*Instructions* : (1) All Questions are *compulsory*.

- (2) Answer each next main Question on a new page.
- (3) Illustrate your answers with neat sketches wherever necessary.
- (4) Figures to the right indicate full marks.
- (5) Assume suitable data, if necessary.
- (6) Use of Non-programmable Electronic Pocket Calculator is permissible.
- (7) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.
- (8) Use of steam tables, logarithmic, Mollier's chart is permitted.

#### Marks

#### **1.** Attempt any TEN of the following :

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- (a) Define organic and inorganic chemistry.
- (b) Give the structure of following organic compound :
  - (i) Ethanoic acid
  - (ii) Formaldehyde
- (c) Give any four physical properties of alkanes.
- (d) Differentiate between Alicyclic compounds and Hetrocyclic compounds.
- (e) Define saturated and unsaturated hydrocarbons.
- (f) Write four physical properties of Alcohol.

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- (g) Give classification of aromatic compound.
- (h) Define functional group. Explain it with suitable examples.
- (i) State Raoult's Law.
- (j) Define indicator and give any two examples.
- (k) Give IUPAC name of
  - (i) n-propyl alcohol
  - (ii) sec-butyl alcohol
- (l) Distinguish between alcohol and phenol with respect to chemical test.

#### 2. Attempt any FOUR of the following :

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- (a) How organic compound are classified ? State example of each.
- (b) How aliphatic compounds are classified ? Explain with structural formula.
- (c) State rules for nomenclature of branched chain hydrocarbons with suitable example. (any four)
- (d) Give IUPAC name of

(i) 
$$CH_{3} - C - CH_{2}Cl$$
  
(i)  $CH_{3} - C - CH_{2}Cl$   
 $CH_{3}$   
(ii)  $CH_{3} - C - CH_{3}$   
(iii)  $CH_{3} - C - CH_{3}$   
(iii)  $CH_{3} - CH - COOH$   
 $NH_{2}$ 

(iv) 
$$CH_2 = CH CH_2 Cl$$

- (e) Describe common system and IUPAC nomenclature methods of organic compounds.
- (f) Give any two reactions of alkanes.

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#### **3.** Attempt any FOUR of the following :

- (a) Give two methods of preparation of alkenes.
- (b) Explain the following terms :
  - (i) Isomerism
  - (ii) Polymerisation
- (c) Explain Baeyer's strain theory.
- (d) Give the uses of Acetylene (any eight).
- (e) Explain wurtz-fitting reaction on Benzene.
- (f) Write following reaction on Benzene :
  - (i) Nitration
  - (ii) Combustion

#### 4. Attempt any FOUR of the following :

- (a) Explain chain Isomerism and position Isomerism.
- (b) How will you prepare benzene
  - (i) by reduction of ketone ?
  - (ii) by action of alkyl halides ?
- (c) Give any two methods of preparation of phenols.
- (d) Differentiate between primary, secondary and tertiary alcohols.
- (e) Explain minimum boiling azeotropes mixture with vapour liquid equilibrium diagram.
- (f) Distinguish between ideal and non-ideal solutions.

#### 5. Attempt any FOUR of the following :

- (a) Define solution and state four types of solution with example.
- (b) What are paraffins ? Why are they called so ?
- (c) Give any two reaction of alkenes.

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- (d) How will you prepare alcohols by
  - (i) Ethyl iodide ?
  - (ii) Methyl bromide ?
- (e) Give the uses of alcohols (any four).
- (f) How vapour pressure of solvent lowered by addition of non-volatile solute ?

#### 6. Attempt any FOUR of the following :

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- (a) Explain Ostwald's theory of Acid-Base indicators.
- (b) Write general formula of alkene & alkyne and state two examples of each.
- (c) Give any four physical properties of Alkyl Halides.
- (d) What are alkyl halides ? How are they classified ?
- (e) Distinguish between monohydric, dihydric phenols and write two uses of phenol.
- (f) Explain Friedal-Craft's reaction. How organic compound prepared by this reaction (any one reaction) ?