

17314

11718

3 Hours / 100 Marks

Seat No.

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- Instructions :**
- (1) All Questions are *compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. (A) Answer any SIX :

12

- (a) Write reactions involved in manufacturing of sulphuric acid.
- (b) State Le-Chatelier's principle.
- (c) Describe biuret and write its chemical formula.
- (d) List any four types of cement.
- (e) Write chemical formula of gypsum.
- (f) Write formula of single super phosphate and triple super phosphate.
- (g) Name the methods for manufacturing of hydrochloric acid.
- (h) Define dry ice. Write its two uses.

(B) Answer any TWO :

8

- (a) Write the notations for diaphragm cell and mercury cell.
- (b) Draw a process flow-sheet for manufacturing of carbon dioxide from flue gas.
- (c) Distinguish between yellow phosphorus and red phosphorus.

2. Answer any TWO : 16

- (a) Explain manufacturing process of sulphuric acid with flow diagram.
- (b) Describe manufacturing process of urea with reactions and flow diagram.
- (c) Explain manufacturing of sodium carbonate by Solvay's process with reactions and flow diagram.

3. Answer any FOUR : 16

- (a) Explain manufacturing of triple super phosphate with reaction.
- (b) Explain manufacturing of phosphorus pentachloride with reaction.
- (c) Distinguish between dry process and wet process in cement manufacturing.
- (d) Write reactions involved in manufacturing of water gas and producer gas.
- (e) Explain manufacturing of Nitrogen and oxygen by Linde's process.
- (f) State properties and uses of sodium hydroxide and chlorine (two each).

4. Answer any FOUR : 16

- (a) Write reactions involved in manufacturing of Ammonium sulphate and Ammonium phosphate.
- (b) Explain hardening and setting of cement.
- (c) Write composition of any two types of cement.
- (d) Describe manufacturing of producer gas.
- (e) Explain manufacturing of acetylene.
- (f) Draw a neat labelled diagram of manufacturing of HCl gas by synthesis process.

5. Answer any TWO :**16**

- (a) Explain manufacturing of Nitric acid with reactions and flow-sheet.
- (b) Explain electric arc process for manufacturing of phosphorus. State uses of phosphorus.
- (c) Explain manufacturing of hydrogen by natural gas with process flow diagram.

6. Answer any FOUR :**16**

- (a) Draw PFD for manufacturing of Ammonia.
 - (b) Explain manufacturing of phosphoric acid by HCl leaching.
 - (c) Explain manufacturing of HCl by salt and sulphuric acid process.
 - (d) Explain working of ammonia converter with neat diagram.
 - (e) Draw process flow diagram for manufacturing of phosphoric acid by sulphuric acid leaching.
 - (f) Explain functioning of carbonating tower and ammoniation tower in manufacturing of soda ash.
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