17211

21819 2 Hours / 50 Marks

1.

Seat No.

Instructions : (1) All Questions are *compulsory*.

- (2) Answer each next main Question on a new page.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data, if necessary.
- (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

Attempt any NINE : 18 Name two ores of copper with their chemical formula. (a) (b) Draw flow chart for the extraction of metal from it's ore. Write chemical reaction of Aluminium with concentrated sulphuric acid. (c) Define : (d) (i) Atmospheric corrosion. Electrochemical corrosion. (ii) (e) Define corrosion. State its types. Define : Cementation (f) (i) Sherardizing (ii) State two characteristics of good paint. (g) Define : Specific Conductance (h) (i) Equivalent conductance (ii)

[1 of 2] P.T.O.

17211

[2 of 2]

- (i) State two applications of photo conductive polymers.
- (j) Distinguish between dielectric and insulator.
- (k) Define : (i) Discharging
 - (ii) Charging
- (l) Write two properties of silicon fluids.

2. Attempt any FOUR :

- (a) Explain smelting process in the extraction of copper metal.
- (b) Explain construction and working of Dry cell.
- (c) Write chemical reactions involved during discharging of Lead acid storage cell.
- (d) Explain purification of Bauxite ore by Bayer's process.
- (e) Write two properties and two applications of Epoxy resign.
- (f) Write composition, properties and applications of tinmann's solder.

3. Attempt any FOUR :

- (a) Explain factors affecting rate of electrochemical corrosion.
- (b) Explain mechanism of electrochemical corrosion with evolution of Hydrogen gas.
- (c) Explain Galvanizing process with neat and labelled diagram.
- (d) Define electrochemical cell. Distinguish between primary cell and secondary cell.
- (e) Explain construction and working of Hydrogen oxygen fuel cell.
- (f) Explain only construction, working and applications of Ni-Cd cell with neat and labelled diagram.

16

16