



MODEL ANSWER
Summer – 17 EXAMINATION

Subject Code:

17407

Important Instructions to examiners:

- 1) The answers should be examined by key words and not as word-to-word as given in the model answer scheme.
- 2) The model answer and the answer written by candidate may vary but the examiner may try to assess the understanding level of the candidate.
- 3) The language errors such as grammatical, spelling errors should not be given more importance. (Not applicable for subject English and Communication Skills).
- 4) While assessing figures, examiner may give credit for principal components indicated in the figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5) Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answers and model answer.
- 6) In case of some questions credit may be given by judgment on part of examiner of relevant answer based on candidate's understanding.
- 7) For programming language papers, credit may be given to any other program based on equivalent concept.

Q. No.	Sub Que.	Answer	Marking Scheme
1	A	Attempt any six of the following	12
1A	a	Define entropy and enthalpy.	2
	Ans.	<p>Entropy : Entropy is defined as a thermodynamic property of a working substance which increase with addition of heat and decreases with removal of heat. Entropy is represented by symbol “S”.</p> <p>Change in entropy can be expresses as</p> $\Delta S = \frac{dQ}{T}$ <p>Where T= absolute temperature dQ= change in heat ΔS= change in entropy.</p> <p>Enthalpy: Enthalpy is defined as the total heat content or total useful energy of a substance. The symbol for enthalpy is “h.” Enthalpy is also considered to be the sum of internal energy “u” and flow energy (or flow work) p.V. This definition of enthalpy can be expressed, mathematically, as follows:</p> $h = u + p.V$ <p>Where, h = Specific enthalpy, measured in kJ/kg (SI Units) u = Specific internal energy, measured in kJ/kg (SI Units) p.V = Flow Energy, Flow Work or p-V work, quantified in kJ/kg (SI Units)</p>	<p>1</p> <p>1</p>



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1.A	e	<p>Plot P-V and T-S diagram for open cycle gas turbine.</p> <p style="text-align: center;">(b) P-v diagram (a) T-s diagram</p> <p style="text-align: center;">Open cycle gas turbine</p>	02
1.A	f	<p>Give classification of renewable Energy sources.</p> <p>Renewable energy sources are classified as: (any four, 02 marks)</p> <ol style="list-style-type: none"> 1. Wind power 2. Biomass 3. Solar applications of energy 4. Hydraulic power 5. Fuel cells 6. Bio fuels 	02
1A	g	<p>List any four merits of liquid fuels over gaseous fuels.</p> <p>Ans. Merits of liquid fuels over gaseous fuels: (For Any four points 02 marks)</p> <ol style="list-style-type: none"> i) Comparatively Less space required to storage. ii) Comparatively Less chance of explosion. iii) There is no loss of heat during storage. iv) Comparatively less inflammable. v) Comparatively easy to handle. vi) Comparatively low cost. 	02
1A	h	<p>Enlist any four types of gaseous fuels.</p> <p>Ans. Types of gaseous fuels: (Any four)</p> <ol style="list-style-type: none"> 1) Natural fuel 2) CNG 3) LPG 4) Water gas 5) Producer gas 6) Coal gas 7) Blast Furnace gas 8) Coke oven gas 9) Oil gas 	02



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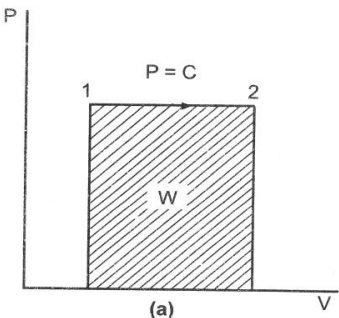
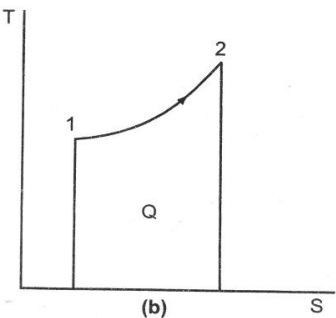
1	B	Attempt any two of the followings.	08																				
1B	a	Represent Carnot cycle on P-V and T-S diagram and write equation for air standard efficiency.	04																				
	Ans.	<p style="text-align: center;">Carnot cycle on P-V and T-S diagram</p> <p>Air standard efficiency of Carnot Cycle:- $\eta = (T_1 - T_2) / T_1$ Where, T₁= temperature of source T₂= temperature of sink</p>	02																				
1B	b	Explain Latent heat and Sensible heat.	04																				
	Ans.	<p>Latent heat: It is defined as the quantity of heat required for phase change of working substance at saturation temperature. OR The amount of heat added at saturation temperature is called latent heat. Latent heat is usually expressed in kJ/kg. Depending upon the change in state which a substance undergoes, different names have been given to this latent heat. They are as given in Table.</p> <table border="1" data-bbox="324 1533 1364 1795"> <thead> <tr> <th>Sr. No.</th> <th>Change of state</th> <th>Heat added/removed</th> <th>Type of latent heat</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>Solid to liquid</td> <td>Added</td> <td>Latent Heat of fusion</td> </tr> <tr> <td>2</td> <td>Liquid to solid</td> <td>Removed</td> <td>Latent heat of solidification</td> </tr> <tr> <td>3</td> <td>Liquid to vapour</td> <td>Added</td> <td>Latent heat of evaporation / vaporization</td> </tr> <tr> <td>4</td> <td>Vapour to liquid</td> <td>Removed</td> <td>Latent heat of condensation</td> </tr> </tbody> </table> <p>Sensible heat: It is defined as the quantity of heat which can be sensed by the thermometer. OR The amount of heat added up to saturation temperature is called sensible heat. The word 'sensible' itself suggest that, it is the heat which can be sensed. As opposed to latent heat, it is the heat responsible for changing the temperature of a substance. Change in temperature can be sensed/perceived and hence it got the name as sensible heat.</p>	Sr. No.	Change of state	Heat added/removed	Type of latent heat	1	Solid to liquid	Added	Latent Heat of fusion	2	Liquid to solid	Removed	Latent heat of solidification	3	Liquid to vapour	Added	Latent heat of evaporation / vaporization	4	Vapour to liquid	Removed	Latent heat of condensation	02
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1B	c	Give classification of gas turbine.	04
	Ans.	Classification of gas turbine: (Note: Any four points.) 1. According to the path of the working substance: i) Open cycle gas turbine ii) Close cycle gas turbine iii) Semi-closed cycle gas turbine 2. According to process of combustion: i) Constant pressure gas turbine ii) Constant volume gas turbine 3. According to direction of flow: i) Radial flow ii) Axial flow iii) Tangential flow 4. According to principle of action of expanding gases: i) Impulse turbine ii) Reaction turbine 5. According to their usage: i) Constant speed ii) Variable speed	
2		Attempt any four of the following:	16
2	a	Plot isobaric process with help of P-V and T-S diagram. Write formula for workdone and Internal Energy.	04
		Isobaric Process  (a) P-V diagram  (b) T-S diagram	02



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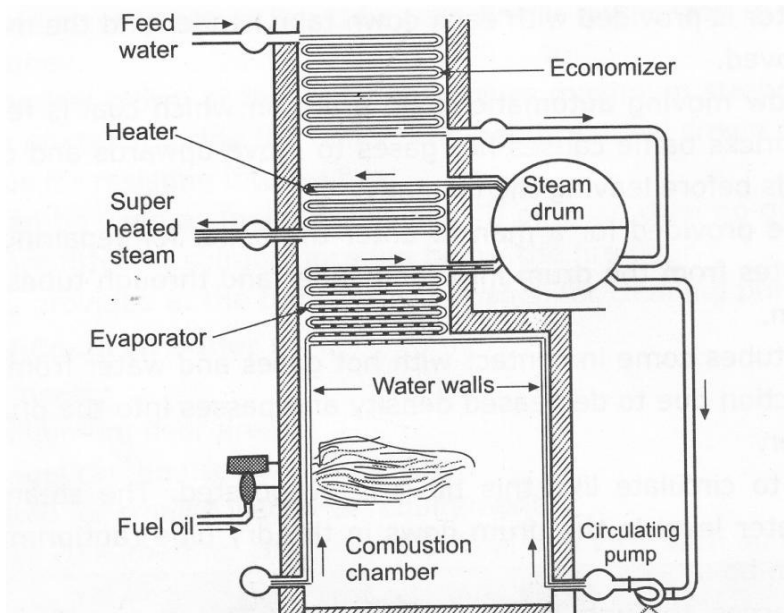
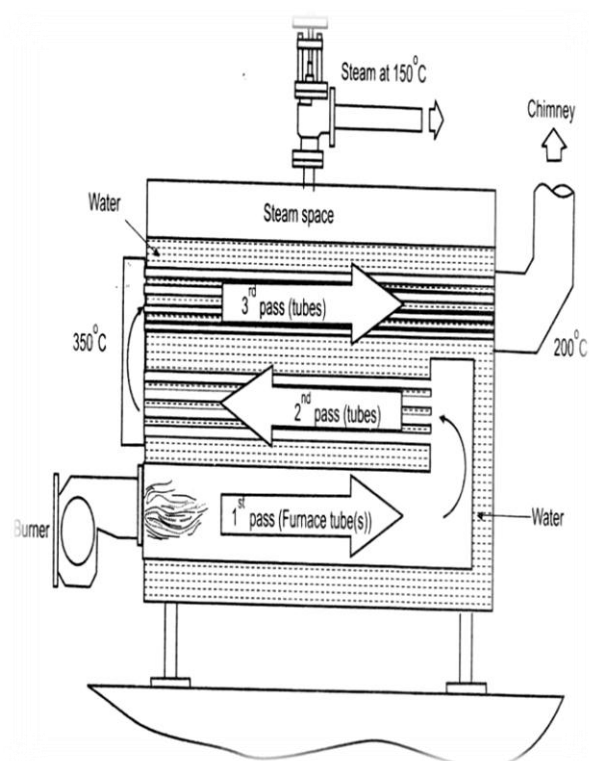
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		<p>Formula for Workdone:</p> $W=P(V_2-V_1)=mR(T_2-T_1)$ <p>Formula for Internal Energy:</p> <p>Change in internal energy during isobaric process is given as</p> $dU= m C_v (T_2-T_1)$	<p>01</p> <p>01</p>																		
2	b	Differentiate between conduction and convection.	04																		
	Ans.	<p>Difference between conduction and convection (for any four point, 4marks)</p> <table border="1"> <thead> <tr> <th>Sr. No.</th> <th>Conduction</th> <th>Convection</th> </tr> </thead> <tbody> <tr> <td>1</td> <td>It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another without displacement of molecules or due to the vibrations of molecules.</td> <td>It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another with displacement of molecules or due to the fluid flowing.</td> </tr> <tr> <td>2</td> <td>It is the mode of heat transfer in which fluid particles do not mix with each other.</td> <td>It is the mode of heat transfer in which fluid particles mix with each other.</td> </tr> <tr> <td>3</td> <td>It occurs in solid.</td> <td>It occurs in liquid and gases.</td> </tr> <tr> <td>4</td> <td>It governs by Fourier's law of heat conduction.</td> <td>It governs by Newton's law of convection heat transfer.</td> </tr> <tr> <td>5</td> <td>Example: Heat flow from one end to other end of metal rod.</td> <td>Example: Heat flow from boiler shell to water.</td> </tr> </tbody> </table>	Sr. No.	Conduction	Convection	1	It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another without displacement of molecules or due to the vibrations of molecules.	It is the mode of heat transfer from one part of substance to another part of same substance or one substance to another with displacement of molecules or due to the fluid flowing.	2	It is the mode of heat transfer in which fluid particles do not mix with each other.	It is the mode of heat transfer in which fluid particles mix with each other.	3	It occurs in solid.	It occurs in liquid and gases.	4	It governs by Fourier's law of heat conduction.	It governs by Newton's law of convection heat transfer.	5	Example: Heat flow from one end to other end of metal rod.	Example: Heat flow from boiler shell to water.	04
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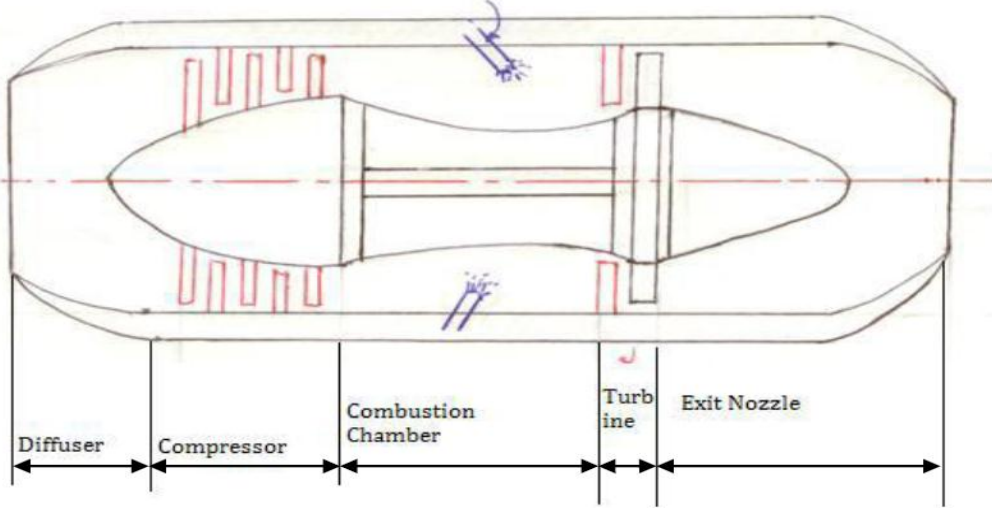
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2	c	Draw neat labeled sketch of La-Mont Boiler.	04
	Ans.	 <p align="center">La-Mont Boiler</p>	04
2	d	Draw neat labeled sketch of three pass packaged type boiler.	04
		 <p align="center">Three pass packaged type boiler</p>	04

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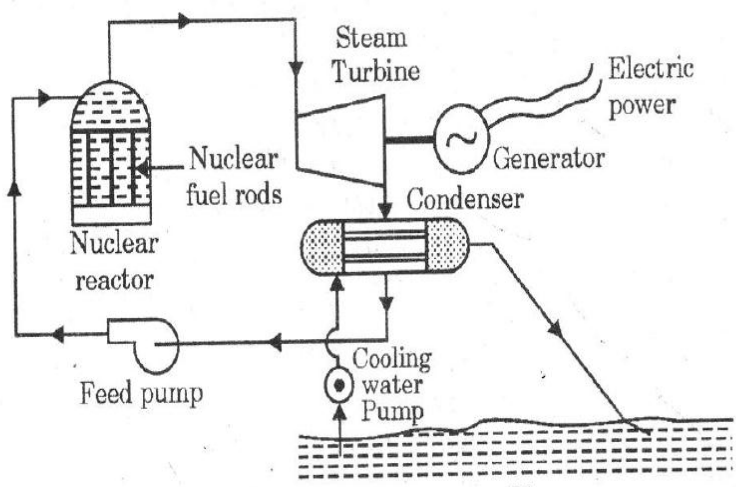
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3	b	Explain with neat sketch turbojet engine.	04
	Ans.	<p>Answer: Turbo-jet Engine:</p> <p>Turbo-jet engine consists of diffuser, compressor, combustion chamber turbine and nozzle. At entrance air diffuser causes rise in pressure in entering air by slowing it down. A rotary compressor, which raises the pressure of air further to required value and delivers to the combustion chamber. The compressor is axial or radial type driven by turbine. In the combustion chamber, fuel is sprayed, as result of this combustion takes place at constant pressure and the temperature of air is raised. Then this product of combustion passes into the gas turbine gets expanded and provides necessary power to drive the compressor. The discharge nozzle in which expansion of gases is completed and thrust of propulsion is produced. The velocity in the nozzle is grater then flight velocity.</p> <div style="text-align: center; margin: 10px 0;">  </div> <p align="center">Fig. Turbo-jet Engine</p>	02
			02

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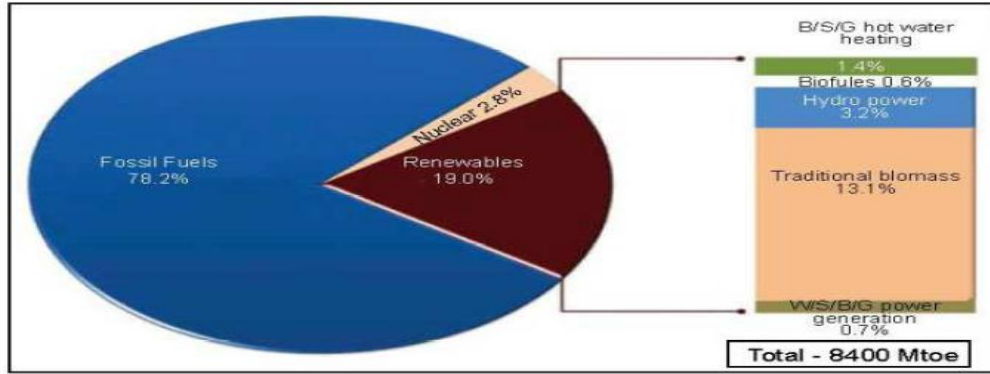
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3	c	Draw neat labeled sketch of nuclear power plant.	04
	Ans.	<p>Answer: Nuclear Power Plant:</p> <div style="text-align: center;">  <p>Fig. Nuclear Power Plant</p> </div>	04
3	d	Explain the importance of non-conventional power generation system in the present situation of power shortage throughout the world.	04
	Ans.	<p>Answer</p> <p>Due to the problems associated with the use of fossil fuels, alternative sources of energy have become important and relevant in today's world. These sources, such as the sun and wind, can never be exhausted and are therefore called renewable. Also known as non-conventional sources of energy, they cause less emission and are available locally. Their use can significantly reduce chemical, radioactive, and thermal pollution. They are viable sources of clean and limitless energy. Most of the renewable sources of energy are fairly non-polluting and considered clean that is why every body is looking for an alternative energy source to improve economy, to meet energy demand, to mitigate pollution, etc., (Manwell et al, 2007). Thus Renewable energy sources have the twin advantages of free availability and non-polluting nature.</p>	04

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Renewable Energy Share in Global Energy Consumption

(Note: W-Wind; B-Biomass; S-Solar; G-Geothermal)

3 e Compare ultimate analysis and proximate analysis of solid fuels. 04

Ans.

Sr. No.	Ultimate analysis	Proximate analysis
01	Ultimate analysis is coal is complete breakdown of coal into chemical constituents	Proximate analysis is coal is complete breakdown of coal into physical constituents
02	This analysis gives percentage of carbon, hydrogen, oxygen, Sulphur and ash.	This analysis gives percentage of moisture, volatile matter, fixed carbon and ash.

04

3 f 04

During a boiler trial the coal analysis on mass basis was reported as :
 $C = 62.4\%$, $H_2 = 4.2\%$, $O_2 = 4.5\%$, Moisture = 15% & ash – 13.9%.
 Calculate minimum air required to burn 1 kg of coal. Also calculate higher and lower calorific value.

Soln : Given

Composition of coal on mass basis.

Carbon (C) = 62.4 = 0.624

Hydrogen (H₂) = 4.2% = 0.042

Oxygen (O₂) = 4.2% = 0.045

Moisture = 15% = 0.15

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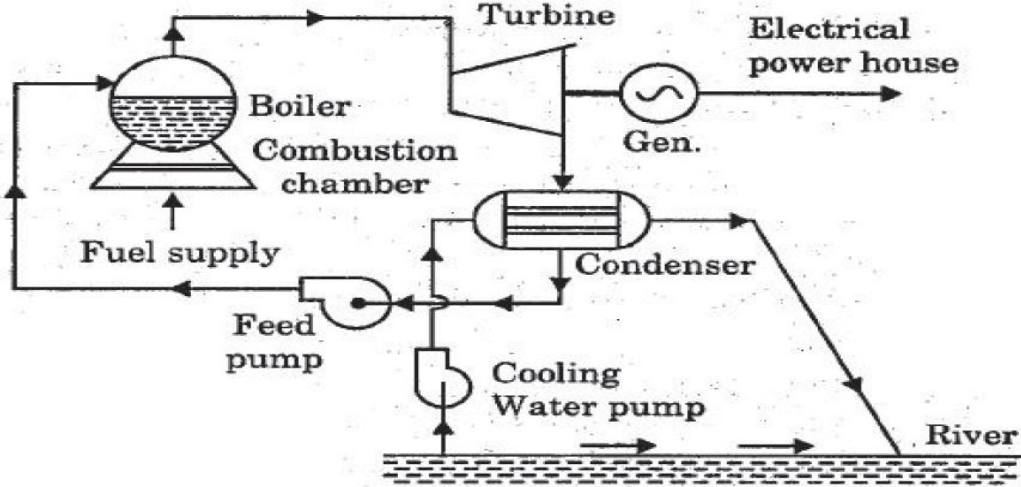
		<p>Ash = 13.9 = 0.139</p> <p>Now, Minimum mass of air required for complete combustion of 1 Kg. of fuel. $= 100/23 (2.67 C + 8 H_2 + S - O_2)$ Kg. $= 100/23 (2.67 \times 0.624 + 8 \times 0.042 + 0 - 0.044)$ $= 100/23 (2.136 + 0.264 + 0.009 - 0.004)$ $= 100/23 (1.666 + 0.336 - 0.045)$ $= 8.613$ Kg. per Kg. of Coal burnt.</p> <p>We know Dulong's formula. H.C.V. of Coal = $33800 C + 144000 (H_2 - O_2/8) + 9270 S$ KJ/Kg. $= (33800 \times 0.624) + [144000 (0.042 - 0.045/8)] + (9270 \times 0)$ H.C.V. = 26329.2 KJ/Kg.</p> <p>L.C.V. of Coal = H.C.V. - $9 H_2 \times 2466$ KJ/Kg. $= 26329.2 - (9 \times 0.042 \times 2466)$ $= 26329.2 - 932.148$ $= 25397.052$ KJ/Kg</p>	<p>01</p> <p>01</p> <p>01</p> <p>01</p>
4		Attempt any TWO of following.	16
4	a	Describe construction of a thermal power plant with neat sketch and explain its working. What are the parameters to be taken into account for site selection of thermal power plant ?	08
Ans.		<p><u>Thermal Power Plant :</u></p> 	02

Figure :- thermal power plant.



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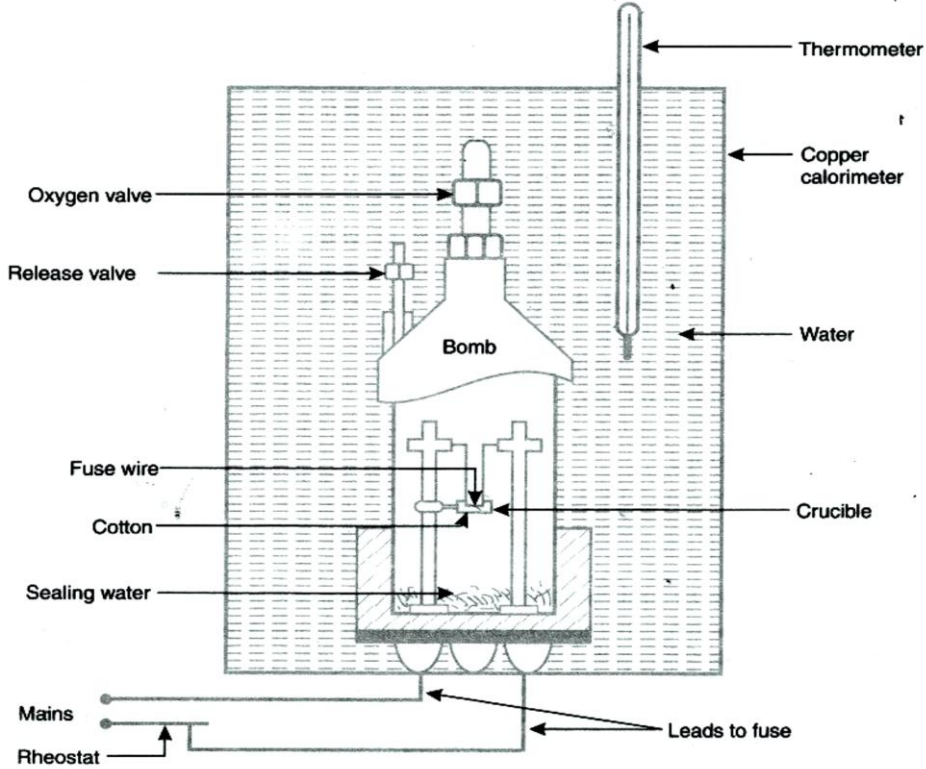
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	<p>In above figure shows major components of thermal power plant which are namely as i) Boiler, ii) Steam turbine, iii) Electric generator, iv) Condenser, v) Combustion chamber, vi) Feed pump etc.</p> <p>Generally for run such type of power plant we can use fuel in form of solid (i.e. coal), liquid (oil) (or) gaseous for the production of steam.</p> <p>Here steam is generated using storage energy in fuel.</p> <p>Initially fuel is supplied into combustion chamber for combustion process. After combustion this heat is given to boiler. Due to this heat water is converted into steam.</p> <p>Now this steam is used to run steam turbine. This steam turbine is directly connected to electrical generator which is used to produce electrical energy.</p> <p>Now the steam coming out of turbine is allowed to pass through condenser in which it is condensed with the help of cooling water.</p> <p>Here condensate is again pumped to boiler for the formation of steam. This type of plant works on closed cycle. Fluid is used again and again for the purpose of power generation.</p> <p>In India Coal is used for run thermal power plant. Oil is used in U.S.A. and gas is used in Canada.</p> <p>Choice of fuel based on availability and economy of country.</p> <p>Site selection for thermal power plant:</p> <ol style="list-style-type: none">1) Availability of land, workers is very important.2) For transportation of fuel power plant should be near to road track (or) rail facility.3) For cooling purpose large amount of water is needed near to power plant.4) Generally in India for run thermal power plant coal is used so power plant should be near to coal mines.5) Ash disposal facility should be available.6) There should be scope for future development.7) Site should be safe and away from urban area.8) Cost of land, infrastructure etc. should be minimum.	<p>03</p> <p>03</p>
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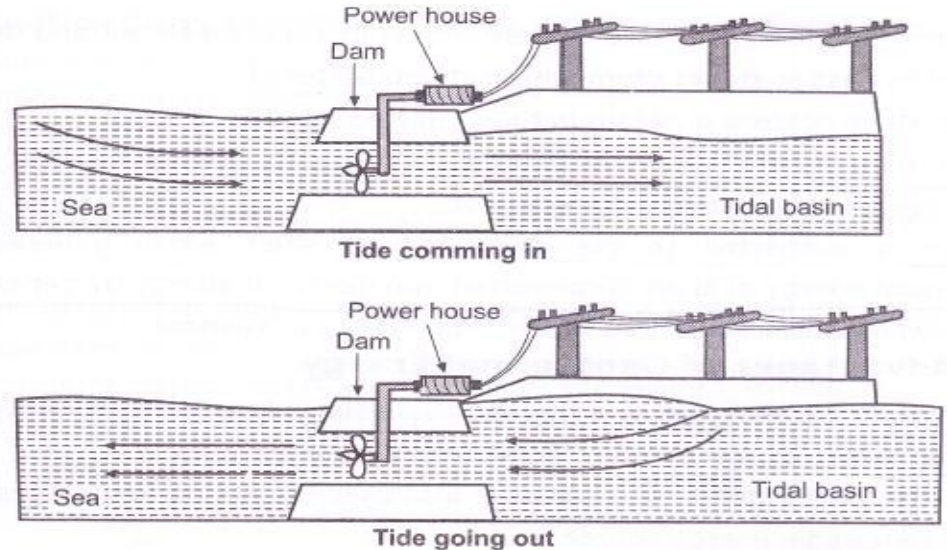
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4	b	Describe with neat sketch construction and working of Bomb calorimeter. Write Dulong's formula and state its use.	08
	Ans.	<p>Bomb calorimeter:</p>  <p>The calorific value of solid and liquid fuels is determined in the laboratory by 'Bomb calorimeter'. It is so named because its shape resembles that of a bomb. Fig. shows the schematic sketch of a bomb calorimeter.</p> <p>Construction : The calorimeter is made of austenitic steel which provides considerable resistance to corrosion and enables it to withstand high pressure. In the calorimeter, a strong cylindrical bomb in which combustion occurs.</p> <p>The bomb has two valves at the top. One supplies oxygen to the bomb and the other releases the exhaust gases. A crucible in which a weighed quantity of fuel sample is burnt is arranged between the two electrodes as shown in fig. The calorimeter is fitted with a water jacket which surrounds the bomb. To reduce the losses due to radiation, the calorimeter is further provided with a jacket of water and air. A stirrer for keeping the</p>	04
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		<p>temperature of water uniform and a thermometer the temperature up to accuracy of 0.0010 C is fitted through the lid of the calorimeter. The heat released by the fuel on combustion is absorbed by the surrounding water and the calorimeter. From the above data the calorific value of the fuel can be found.</p> <p>Dulong's formula used to calculate the theoretical calorific value of fuel if ultimate analysis is available and the calorific value of elementary combustibles are known.</p> $\text{Theoretical calorific Value of fuel} = 33800 C + 144500 \left(H_2 - \frac{O_2}{8} \right) + 9300 S \text{ kJ/kg}$ <p>Where C, H₂ O₂ & S repents the mass of carbon, hydrogen, oxygen and sulfur in kJ/Kg</p>	02
4	C	Attempt the following:	08
4	C(i)	(i) Explain the tidal power plant.	04
Ans.	(i) Working of Tidal power plant:	<p>During high tide the water flow from sea into the tidal basin through water turbine as the level of water in sea is more than tidal basin. This operates the turbine and generator and power is produced. Potential energy of sea water converted into mechanical energy by turbine and it converts into electrical by generators. During low tide water flow from tidal basin into sea as water level in the sea is lower than basin level in both cases generation of power is same. Only difference in that rotation of turbine blade is opposite.</p>	02
			02



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	<p>Heat supplied = $Q_a = C_p (T_3 - T_2)$</p> <p>Heat rejected = $Q_r = C_v (T_4 - T_1)$</p> <p>Work done = $Q_a - Q_r$</p> $= C_p (T_3 - T_2) - C_v (T_4 - T_1)$ <p>Air standard efficiency is given by,</p> $\eta_{ase} = \frac{\text{Net work done}}{\text{Heat addition}}$ $= \frac{C_p(T_3 - T_2) - C_v(T_4 - T_1)}{C_p(T_3 - T_2)}$ $= 1 - \frac{C_p}{C_v} \frac{T_4 - T_1}{(T_3 - T_2)}$ $\eta_{ase} = 1 - \frac{T_4 - T_1}{\gamma(T_3 - T_2)} \text{ ----- (1)}$ <p>Consider isentropic process 1-2,</p> $\frac{T_2}{T_1} = \left(\frac{V_1}{V_2}\right)^{\gamma-1}$ $T_2 = T_1 \left(\frac{V_1}{V_2}\right)^{\gamma-1} = T_1 (r)^{\gamma-1}$ <p>Consider constant pressure heat addition process 2-3,</p> <p>As per Charle's law,</p> $\frac{V}{T} = C$ $\frac{V_2}{T_2} = \frac{V_3}{T_3}$ $T_3 = \frac{V_3}{V_2} \times T_2 = \rho \times T_2$ <p>Consider isentropic process 3-4,</p> $\frac{T_4}{T_3} = \left(\frac{V_3}{V_4}\right)^{\gamma-1}$	<p>01</p> <p>01</p> <p>01</p> <p>01</p>
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		$T_4 = T_3 \left(\frac{V_3}{V_2} \times \frac{V_2}{V_4} \right)^{\gamma-1}$ $T_4 = T_3 \left(\frac{\rho}{r} \right)^{\gamma-1}$ $T_4 = \rho \times T_1 (r)^{\gamma-1} \left(\frac{\rho}{r} \right)^{\gamma-1} = T_1 \rho^\gamma$ <p>Substituting the value of T_2, T_3 and T_4 in equation 1,</p> $\eta_{ase} = 1 - \frac{T_4 - T_1}{\gamma(T_3 - T_2)}$ $\eta_{ase} = 1 - \frac{1}{\gamma} \left[\frac{T_1 \rho^\gamma - T_1}{\rho T_1 (r)^{\gamma-1} - T_1 (r)^{\gamma-1}} \right]$ $= 1 - \frac{1}{\gamma} \left[\frac{T_1 (\rho^\gamma - 1)}{T_1 (r)^{\gamma-1} (\rho - 1)} \right]$ $\eta_{ase} = 1 - \frac{1}{(r)^{\gamma-1}} \left[\frac{(\rho^\gamma - 1)}{\gamma(\rho - 1)} \right]$	02
5	(b)	<p>Determine the quantity of heat required to produce 1 kg of steam at a pressure of 6 bar at a temperature of 250°C, under the following conditions:</p> <p>i) When steam is the wet having a dryness fraction 0.9. ii) When steam is dry saturated. iii) When it is superheated at a constant pressure at 250°C. (Take $C_p = 2.3$ KJ/kgK For $P = 6$ bar, $h_f = 670.4$ KJ/kg, and $h_{fg} = 2085$ KJ/kg and $t = 158.8^\circ\text{C}$)</p>	08
		<p>Answer: Given: $P = 6$ bar $x = 0.9$ $\Delta_T = 25^\circ\text{C}$ $C_{pw} = 4.187$ KJ/kgK $C_{psup} = 2.3$ KJ/kgK $T_{sup} = 250^\circ\text{C}$ $T_{sat} = 158.8^\circ\text{C}$ $m = 1$ Kg From steam table, At 6 bar $h_f = 670.4$ KJ/kg, and $h_{fg} = 2085$ KJ/kg</p> <p>i) Wet steam $h_{wet} = h_f + x \cdot h_{fg}$ $= 670.4 + 0.9 \times 2085$</p>	



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	<p>$h_{wet} = 2546.9 \text{ KJ/kg}$</p> <p>Heat already present in water $= m C_{pw} \Delta T$</p> $= 1 \times 4.187 \times 25$ $= 104.675 \text{ KJ/Kg}$ <p>Heat required per Kg of steam $= h_{wet} - \text{Heat already present in water}$</p> $= 2546.9 - 104.675$ $= 2442.225 \text{ KJ/Kg or KJ}$ <p>ii) Dry saturated steam</p> $h_g = h_f + h_{fg}$ $= 670.4 + 2085$ $h_g = 2755.4 \text{ KJ/kg}$ <p>Heat required per Kg of steam $= h_g - \text{Heat already present in water}$</p> $= 2755.4 - 104.675$ $= 2650.725 \text{ KJ/Kg or KJ}$ <p>iii) Superheated steam</p> $h_{sup} = h_f + h_{fg} + m C_{psup} (T_{sup} - T_{sat})$ $= 670.4 + 2085 + 1 \times 2.3 \times (250 - 158.8)$ $h_{sup} = 2965.16 \text{ KJ/Kg}$ <p>Heat required per Kg of steam $= h_{sup} - \text{Heat already present in water}$</p> $= 2965.16 - 104.675$ $= 2860.485 \text{ KJ/Kg or KJ}$	<p>01</p> <p>01</p> <p>01</p> <p>01</p> <p>01</p> <p>02</p> <p>01</p>
(c)	<p>Explain construction and working of screw compressor. Differentiate between centrifugal and axial flow compressor.</p>	08
	<p>Answer: Note- credit should be given to equivalent figure.</p> <p>Screw compressor:</p> <p>Construction: It consists of two mutually engaged helical grooved rotors which are suitably housed in a casing. Out of two rotors male rotor is driver and female rotor is a driven. Male rotor has four lobes and female rotor as six flutes.</p> <p>Working: During rotation of rotor, air enters and takes space between male and female rotor. This air traps and moves axially and radially with rotation of rotors and gets compressed due to volume reduction. Then this air discharged from upward direction. Speed of rotors is different due to different number of lobes and flutes. It handles 3.5 to 300 m³/min and maximum pressure ratio of 20. This system requires lubrication. This compressor is noisy in operation. Used in refrigeration industry.</p>	<p>01</p> <p>01</p>

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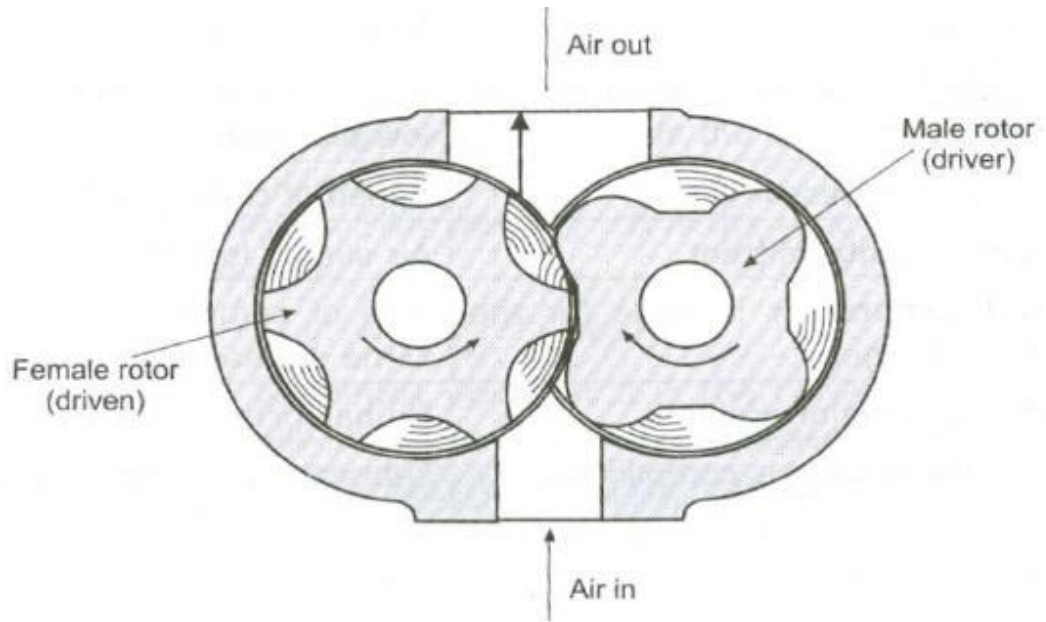


Fig. Screw Compressor

Difference between centrifugal and axial flow compressor (any four points):

Sr. No.	Centrifugal compressor	Axial flow compressor
1	Flow is radial	Flow of air axial
2	Multistaging is difficult	Multistaging is simple
3	Requires low starting torque	Requires high starting torque
4	It is having larger frontal area	It is having less frontal area
5	Isentropic efficiency is 70%	Isentropic efficiency is 80%
6	Low manufacturing and running cost	High manufacturing and running cost
7	This is more compact	This is less compact
8	Pressure ratio per stage is 4:1	Pressure ratio is 1.1 to 1.2
9	It is used in turbojet engine, refrigeration cycle .	It is used in gas turbine power plants.
10	Construction is simple	Construction is complicated

02

04

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		<p>Equation of thermal efficiency (any one):</p> $\eta_{th} = 1 - \frac{T_1}{T_2}$ <p>or</p> $\eta_{th} = \frac{T_2 - T_1}{T_2}$ <p>or</p> $\eta_{th} = 1 - \frac{1}{(r_p)^{\frac{\gamma-1}{\gamma}}}$	01
6	f	What is multi-staging? State necessity of multi-staging and intercooling of compressors.	04
	Ans.	<p>Answer:</p> <p>Multi-staging – For producing high pressure two or more cylinders are provided in series with intercooling arrangement between them. Such an arrangement is known as multi-staging</p> <p>Necessity of multistaging – For producing high pressure i.e. more than 8 bar, single stage air compressor suffers following drawbacks-</p> <ul style="list-style-type: none">i) Size of cylinder is too largeii) Rise in temperature of air is very high. <p>To avoid this difficulty multi-staging is necessary.</p> <p>Necessity of intercooling – In two stage air compressor air is compressed in first cylinder and the temperature of air is increased. If this high temperature air is not passed through intercooler and sent directly to second stage then because of high temperature volume of air increases so amount of air taken inside decreases and pressure is also automatically decreased and volumetric efficiency is also decreases. To avoid this intercooling is necessary.</p>	01 01 02