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16172 3 Hours / 100 Marks

Seat No.

Instructions : (1) All Questions are *compulsory*.

- (2) Answer each next main Question on a new page.
- (3) Illustrate your answers with neat sketches wherever necessary.
- (4) Figures to the right indicate full marks.
- (5) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

 $10 \times 2 = 20$

1. Answer any TEN :

- (a) List the products obtained in coaltar distillation and draw their structures.
- (b) State four physical properties of benzene.
- (c) Complete following reaction and write name of reactant, reagent and product :



- (d) State uses of benzene Sulphonic acid.
- (e) Write the names of following aromatic compounds :



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- (f) State two uses of Aniline.
- (g) How will you prepare benzene diazonium chloride in laboratory ? Write the chemical reactions.
- (h) State physical properties of benzene diazonium chloride.
- (i) Draw the structure of :
 - (i) quinol
 - (ii) resorcinol
- (j) How will you prepare benzoic acid from toluene ?
- (k) Complete the following reaction and predict the product :



- (m) State uses of phenol.
- (n) Explain physical properties of naphthalene.

2. Answer any FOUR :

(1)

- (a) Distinguish between aliphatic and aromatic compounds.
- (b) How will you prepare chlorobenzene using
 - (i) Phenol, and
 - (ii) Cuprous chloride ?

Explain with chemical reactions.

 $4 \times 4 = 16$

 $H^+/H_2O \rightarrow$

- (c) Explain following chemical properties of benzene sulphonic acid :
 - (i) Reaction with NaOH
 - (ii) Reaction with PCl_5
- (d) Name all nitro derivatives which are formed during nitration of benzene. Draw their structures.
- (e) Predict the products of following reactions :

$$\bigvee NH_2 + CH_3COCl \longrightarrow$$
$$\bigvee NH_2 + acetic anhydride \longrightarrow$$

(f) How will you prepare naphthalene and anthracene ? Explain with chemical equations.

3. Answer any FOUR :

$4 \times 4 = 16$

- (a) Explain Friedal Craft's reaction considering toluene as reacting substance.
- (b) What will happen if chlorobenzene reacts with :
 - (i) NH₃
 - (ii) Mg metal
 - (iii) KOH
 - (iv) HNO₃

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(c) Complete following reactions. Draw chemical structures and name reactant, reagent and products :

(i) Benzene sulphonic acid +?
$$\xrightarrow{\text{hydrolysis}}$$
? +?

(ii) Benzene sulphonic acid +? Bromination ? +?

- (d) Explain following chemical properties of nitrobenzene :
 - (i) Reduction in acidic medium and neutral medium
 - (ii) Sulphonation
- (e) State and explain physical properties of aniline.
- (f) Predict the products of following chemical reactions and complete it :



4. Answer any FOUR :

 $4 \times 4 = 16$

- (a) What is coaltar ? How will you get benzene toluene and xylene from coaltar ?
- (b) How will you prepare (any two) :
 - (i) Phenol
 - (ii) Chlorobenzene
 - (iii) Thiophene from benzene diazonium chloride?

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- (c) Explain following chemical reactions of aniline :
 - (i) Oxidation
 - (ii) halogenation
- (d) Why is phenol acidic in nature ? Explain with chemical reactions.
- (e) Complete following chemical reactions and draw the structure where required :



- (f) Explain following chemical reactions :
 - (i) Preparation of β naphthol
 - (ii) oxidation of anthracene

5. Answer any FOUR :

- (a) How will you obtain (any two) :
 - (i) nitrobenzene
 - (ii) benzene sulphonic acid
 - (iii) toluene
 - (iv) acetophenone from benzene?
- (b) Explain following chemical properties (any two) of benzene diazonium chloride :
 - (i) Reduction in presence of $H_3PO_2 + H_2O$
 - (ii) Reaction with KI
 - (iii) Reaction with HBF₄

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- (c) (i) Write common name and chemical structure of hydroxy benzene.
 - (ii) It discolours in air.Explain why.
- (d) How will you obtain (i) ethyl benzoate and (ii) benzoyl chloride from benzoic acid ?
- (e) (i) Write two physical properties of anthracene.
 - (ii) State its applications.
- (f) Naphthalene is used for preparation of dye intermediate. Explain giving suitable chemical reaction.

6. Answer any FOUR :

- (a) Predict the products giving suitable chemical reaction when toluene is treated with :
 - (i) HNO₃, in presence of H_2SO_4 at 160 °C
 - (ii) CH₃C*l*
 - (iii) H_2SO_4
- (b) How will you prepare (any two) :
 - (i) anisole
 - (ii) phenyl hydrazine
 - (iii) p-hydroxy azo benzene

from benzene diazonium chloride ?

- (c) Explain any two chemical properties of phenol :
 - (i) Reaction with HNO₃
 - (ii) Reaction with NaOH followed by $CHCl_3$
 - (iii) Kolbe reaction

$4 \times 4 = 16$

- (d) Explain acidic nature of benzoic acid giving chemical reactions.
- (e) Complete following chemical reactions :



Write the name of products.

(f) Anthracene is used for preparation of dye. Explain giving suitable chemical equation.

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