

17221

16172

3 Hours / 100 Marks

Seat No.

--	--	--	--	--	--	--	--	--

- Instructions :**
- (1) All Questions are *compulsory*.
 - (2) Answer each next main Question on a new page.
 - (3) Illustrate your answers with neat sketches wherever necessary.
 - (4) Figures to the right indicate full marks.
 - (5) Assume suitable data, if necessary.
 - (6) Mobile Phone, Pager and any other Electronic Communication devices are not permissible in Examination Hall.

Marks

1. Solve any FIVE :

20

- (a) Give the classification of organic compounds on the basis of their functional group.
- (b) Explain the mechanism of SN_1 reaction.
- (c) How is ethylene prepared by dehydration of alcohols. Give 2 physical properties & two uses of ethylene.
- (d) Define alcohols. Give it's example. Explain any one method for preparation of alcohols.
- (e) Explain the following reaction with respect to formaldehyde :
 - (i) Addition reaction
 - (ii) Substitution
- (f) What is the action of Heat & KOH on oxalic acid ?
- (g) What are proteins ? Give its classification.

2. Attempt any TWO :**16**

- (a) What are alkanes ? Explain the wurtz synthesis method for preparation of Ethane. How does chlorination of Methane takes place ?
- (b) How is Acetic acid prepared from :
- (i) Grignard Reagent
 - (ii) Alkyl cyanide
- (c) What are amino acids ? Give its classification with suitable examples.

3. Attempt any TWO :**16**

- (a) How is dimethyl ketone prepared from –
- (i) Isopropyl alcohol (ii) Acetic acid. Give four uses of acetone.
- (b) Explain Homologous series with suitable example. What are organic compounds ?
- (c) What is the action of following on methyl alcohol :
- (i) Na-metal (ii) PCl_3 (iii) $SOCl_2$ (iv) H_2SO_4 (20%)

4. Attempt any TWO :**16**

- (a) Explain the following reactions of carboxylic acid with suitable e.g.
- (i) Amide formation (ii) Acid chloride formation.
- (b) Explain the preparation of Ethyne from –
- (i) de-hydrohalogenation (ii) Metallic calcium carbide.
- (c) Describe the method of separating proteins & define isoelectric point.

5. Attempt any TWO :**16**

- (a) Explain the SN_2 mechanism of Nucleophilic substitution reaction.
- (b) Explain addition and elimination reaction with suitable example.
- (c) How is ethylene prepared by –
- dehydration of alcohols
 - thermal & catalytic cracking

6. Attempt any TWO :**16**

- (a) Explain the structural formula of Methane & Ethane
Give the IUPAC nomenclature of following :

- (i)
- $$\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{CH} - \text{CH}_2 - \text{CH}_3 \end{array}$$
- (ii)
- $$\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{CH}_2 - \text{CH} - \text{CH} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$$
- (iii)
- $$\begin{array}{c} \text{CH}_2 - \text{CH}_3 \\ | \\ \text{CH}_3 - \text{CH} - \text{CH}_2 - \text{CH} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$$
- (iv)
- $$\begin{array}{c} \text{CH}_3 \\ | \\ \text{CH}_3 - \text{C} - \text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$$

- (b) How is acetone prepared from –
- acetylene
 - Isopropyl alcohol.
- (c) Distinguish between primary, secondary & tertiary alcohol.
-

