# 17227

# 15162 2 Hours / 50 Marks

Seat No.

### *Instructions* : (1) All Questions are *compulsory*.

- (2) Answer each next main Question on a new page.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data, if necessary.

#### 1. Answer any SEVEN :

- (a) Define : (i) rust, (ii) corrosion.
- (b) Explain why metals are electroplated. Give examples.
- (c) State two prime functions of points.
- (d) Define : (i) Gross calorific value, (ii) net calorific value.
- (e) State significance of proximate analysis.
- (f) Explain how water acts as a lubricant.
- (g) State uses of silicone oils as lubricants.
- (h) Define : (i) oildag (ii) aquadag.
- (i) Define a buffer. Name types of buffer.
- (j) Define : (i) B.O.D., (ii) C.O.D.

#### 2. Answer any FOUR :

- (a) Name types of oxide films, formed on corrosion of metal. Which of them gives best protection against corrosion ? Explain.
- (b) Compare : tinning and galvanising.
- (c) Write procedure for estimation of 'fixed carbon' in solid fuel.
- (d) Define 'biogas'. State its advantages.
- (e) Write selection criteria of lubricants for I.C. engine. Name a suitable lubricant.
- (f) Explain principle of estimation of chlorides in water.

[1 of 2]

Marks

#### $7 \times 2 = 14$

#### $4 \times 3 = 12$

# [2 of 2]

#### 3. Answer any FOUR :

- (a) Describe two methods of protection of metal from corrosion.
- (b) Write characteristics of a good fuel.
- (c) Write typical boiling range, composition and application of gasoline, heavy oil.
- (d) Compare : Flash point and fire point.
- (e) Outline ozonisation method of disinfection of water.
- (f) Compare zeolite process and ion exchange process used to soften hard water.

## 4. Attempt any FOUR :

- (a) Explain mechanism of immersed corrosion.
- (b) Draw Bomb's calorimeter. Explain its working.
- (c) Write main constituents of LPG. State disadvantages of LPG over gasoline as a gaseous fuel.
- (d) Describe with a diagram, boundary lubrication.
- (e) Outline 'reverse osmosis' method of softon water.
- (f) Explain why the use of two metals be avoided.

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#### $4 \times 3 = 12$

 $3 \times 4 = 12$