## 21415 2 Hours / 50 Marks

Seat No.
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**Instructions**: (1) All Questions are *compulsory*.

- (2) Answer each next main Question on a new page.
- (3) Illustrate your answers with neat sketches wherever necessary.
- (4) Figures to the right indicate full marks.

**Marks** 

## 1. Attempt any NINE:

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- (a) Define:
  - (i) Flux
  - (ii) Gangue
- (b) Differentiate between pig iron and cast iron (any 2 points).
- (c) Define heat treatment of steel.
- (d) Why gold does not get corroded in air?
- (e) Write any two advantages of metal spraying.
- (f) State the constituents of paints.
- (g) Why galvanized containers are not used to store food stuffs?
- (h) Define scales and sludges.
- (i) Explain chlorination of water by using chlorine gas.
- (j) Define sterilization and sedimentation.
- (k) List any four constituents of cement.
- (l) What is slaking of lime?

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2	Attempt any FOUR:	
4.	Attempt any FOOK:	

- (a) Write the chemical reactions occurring during conversion of iron oxide to iron in reduction zone of blast furnace at various temperature ranges.
- (b) What is the effect of alloying elements Ni and Co on the properties of steel?
- (c) Differentiate between annealing-normalising.
- (d) Explain stepwise the mechanism of electrochemical corrosion with evolution of hydrogen, along with neat labelled diagram.
- (e) State and explain any two factors affecting rate of atmospheric corrosion.
- (f) Differentiate between Galvanising-Tining.

## 3. Attempt any FOUR:

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- (a) State any adverse effects of using hard water in (1) paper industry (2) sugar industry.
- (b) Name types of impurities present in water. Explain sterilisation of water by using bleaching powder.
- (c) Define:
  - (i) Hard water
  - (ii) Soft water
  - (iii) Degree of hardness
  - (iv) Filtration
- (d) Draw the sketch of reverse osmosis cell and label the parts. Explain desalination of sea water by reverse osmosis.
- (e) Explain permutit method for removal of hardness of water with diagram.
- (f) How does setting and hardening of cement takes place?