

Scheme – I
Sample Question Paper

Program Name : Mechanical Engineering Program Group
Program Code : ME / PG / PT
Semester : Third
Course Title : Thermal Engineering
Marks : 70

22337

Time: 3 Hrs.

Instructions:

- (1) All questions are compulsory.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data if necessary.
- (5) Preferably, write the answers in sequential order.
- (6) Use of steam table and Mollier chart is permitted.

Q.1) Attempt any FIVE of the following.

10 Marks

- (a) Explain similarities between Heat & Work.
- (b) State Avogadro's law.
- (c) Define sensible heat and latent heat
- (d) Define bleeding and regenerative feed heating.
- (e) Explain the purpose of governing of steam turbines.
- (f) Define Mach number and critical pressure.
- (g) Define thermal Conductivity. State it's Unit.

Q.2) Attempt any THREE of the following.

12 Marks

- (a) Apply steady state energy equation with a block diagram to Boiler & IC Engine
- (b) Define isothermal process and plot it on Pv & Ts diagrams.
- (c) Describe generation of steam at constant pressure with temperature enthalpy diagram
- (d) Define enthalpy of evaporation, dryness fraction, degree of superheat

Q.3) Attempt any THREE of the following.

12 Marks

- (a) Explain different losses in steam turbines
- (b) Explain principle of working of reaction steam turbine with a sketch.
- (c) Air is expanded from 7 bar and 80 °C to 1 bar according to law $PV^{1.2} = C$. Plot the process on Pv & Ts diagram and state the formula to be used to find out the work done during the above process.
- (d) Steam expands from 20 bar and 300 °C to 1 bar at constant entropy. Using mollier chart, find change in enthalpy during the process.

Q.4) Attempt any THREE of the following.

12 Marks

- (a) Differentiate surface and jet condensers (any four points).

- (b) A steel flask of 0.04 m^3 capacity is to be used to store nitrogen at 120 bar and 20°C . The flask is protected against excessive pressure by a fusible plug which will melt and allow gas to escape if temperature rise is too high.
- Estimate mass of nitrogen in kg, the flask will hold at designed condition?
 - At what temperature fusible plug will melt in order to limit pressure of flask to a maximum of 150 bar. Take $R = 287 \text{ J/Kg K}$ (2)
- (c) 1 kg of air at a pressure of 8 bar and a temperature of 100°C undergoes a reversible polytropic process following the law $Pv^{1.2} = C$. If final pressure is 1.8 bar, find final specific volume and temperature. Use $R = 287 \text{ J/Kg K}$ and $\gamma = 1.4$ for air (2)
- (d) Draw a labeled sketch of plate heat exchanger. (6)
- (e) In the given formula for a steam condenser, state meanings of different terms involved with their units.

$$\eta_{vacuum} = \frac{P_{baro} - P_{abs}}{P_{baro} - P_{sat}}$$

Q.5) Attempt any TWO of the following.

12 Marks

- (a) Attempt the following
- Suggest measure to improve thermal efficiency of steam turbine power plant and also discuss limitations for the same.
 - Suggest method for governing of reaction turbines and also discuss limitations for the same.
- (b) A thermo pane window consists of two 5 mm thick glass ($k = 0.78 \text{ W / m K}$) sheets separated by 10 mm stagnant air gap ($k = 0.025 \text{ W / m K}$). The convection heat transfer coefficient for inner and outer air are $10 \text{ W / m}^2 \text{ K}$ and $50 \text{ W / m}^2 \text{ K}$. Determine rate of heat loss per m^2 of the glass surface for a temperature difference of 60°C between inside and outside air. Compare results of heat loss if window has only single sheet of glass of thickness 5 mm instead of thermo pane.
- (c) Attempt the following
- A steam sample was taken at certain point. The mass of water was found 90% more than mass of dry steam. Find dryness fraction.
 - If above steam is at 5 bar find enthalpy required to make it dry saturated.

Q.6) Attempt any TWO of the following.

12 Marks

- (a) Attempt the following
- The enthalpy across a device decreases by 2000 kJ/Kg and the heat transfer from the device is 50 kJ/Kg . Find the power developed assuming device a steam turbine. Take mass flow rate of steam as 1.3 kg/s and neglect kinetic energy changes.
 - The room temperature was found to be maintained at 18°C when atmosphere was at 36°C . Find the Coefficient of Performance of air-conditioner used assuming it is working on carnot cycle.
- (b) Following data is available for a power plant
- Steam Power Plant of 250 MW capacity.
 - Water handled by condenser is large.

iii. Water available for cooling is saline water.

Select a condenser to be required in above situations with justification

- (c) A typical application is to be designed to maintain $-15\text{ }^{\circ}\text{C}$ in a chamber. The atmosphere is at $29\text{ }^{\circ}\text{C}$. Thermal conductivity values for different materials used are: *Steel* – 50 W/mK , *Brass* – 125 W/mK . Select material from above which will give least thickness required for unit heat flow per unit area. Use electrical analogy to solve the problem.

Scheme – I
Sample Test Paper - I

Program Name : Mechanical Engineering Program Group
Program Code : ME/PT/Fabrication Technology
Semester : Third
Course Title : Thermal Engineering
Marks : 20

22337

Time: 1 Hour

Instructions:

- (1) All questions are compulsory.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data if necessary.
- (5) Preferably, write the answers in sequential order.

Q.1) Attempt any FOUR of the following.

08 Marks

- a) State Zeroth Law of Thermodynamics. Give application of the same
- b) State assumptions for Steady State Energy Equation.
- c) What is Universal gas constant? State it's Value with unit.
- d) Define adiabatic process and polytropic process.
- e) Differentiate refrigerator with heat pump.
- f) Write characteristic gas equation & state meaning of each term involved write unit for each term.

Q.2) Attempt any THREE of the following.

12 Marks

- a) Describe with examples flow work and non flow work.
- b) Find enthalpy change when 3 kg of air expands at constant pressure. Initial conditions of air are 1 m^3 & 30°C . Final conditions of air are 1.5 m^3 and 80°C .
- c) Sketch Pv & Ts diagrams for constant volume process.
- d) Find characteristic gas constants for CO_2 , N_2 , O_2 . Value of Universal gas constant is $8.314 \text{ kJ/Kg mole K}$. Values of atomic weights are C – 12, O – 16, N – 14.

Scheme – I
Sample Test Paper - II

Program Name : Mechanical Engineering Program Group
Program Code : ME/PT/Fabrication Technology
Semester : Third
Course Title : Thermal Engineering
Marks : 20

22337

Time: 1 Hour

Instructions:

- (1) All questions are compulsory.
- (2) Illustrate your answers with neat sketches wherever necessary.
- (3) Figures to the right indicate full marks.
- (4) Assume suitable data if necessary.
- (5) Preferably, write the answers in sequential order.
- (6) Use of steam table and Mollier chart is allowed.

Q.1) Attempt any FOUR of the following.

08 Marks

- a) Define Dryness fraction, wet steam.
- b) Select nozzles for different types of turbines.
- c) State need of condenser in a power plant.
- d) State Fourier's law of conduction.
- e) Differentiate mountings and accessories.
- f) Define boiler draught.

Q.2) Attempt any THREE of the following.

12 Marks

- a) Sketch any one high pressure, water tube boiler and label it.
- b) Differentiate between impulse and reaction turbine.
- c) Explain 1. Governing 2. Compounding in steam turbines.
- d) Temperature rise in cooling water of a steam condenser was found 50 °C. Inlet temperature of water is 30 °C. Find condenser efficiency if absolute pressure in condenser is 60 mm of Hg.